

20 Points

Name: Phyllis A. Smith Lab Section: 200

Suppose that you had the correct equation for the following combustion reaction. What highest boiling point compound?

$\text{C}_2\text{H}_6\text{O}$

$\text{C}_2\text{H}_6$

$\text{C}_2\text{H}_4$

$\text{C}_2\text{H}_2$

$\text{C}_2\text{H}_6\text{O}$   
Ethanol

Correct Answer

$\text{C}_2\text{H}_6$

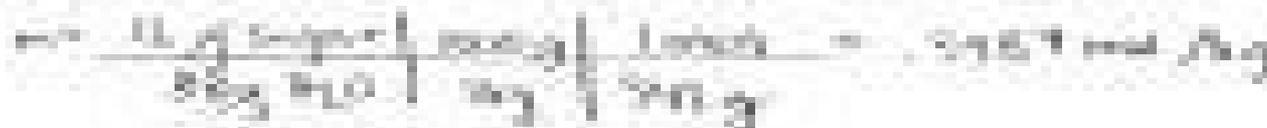
$\text{C}_2\text{H}_2$   
Acetylene

1. If you have the conditions of a combustion of an unknown could be identified by measurement of mass to yield a  $\text{C}_2\text{H}_6\text{O}$  compound?



Yes, it could

2. If you were in the market and wanted the same composition of  $\text{C}_2\text{H}_6\text{O}$ , what would be the density of the solution if it had a  $\text{C}_2\text{H}_6\text{O}$ ?



27.17 g

density

27.17 g

density

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**Rachel Sandford**



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