

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	Very	Very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (III) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Chemistry Acids And Bases D Answers

Scott C. Dulebohn



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Table of Contents Chemistry Acids And Bases D Answers

1. Understanding the eBook Chemistry Acids And Bases D Answers
 - The Rise of Digital Reading Chemistry Acids And Bases D Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Acids And Bases D Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Acids And Bases D Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Acids And Bases D Answers

- Personalized Recommendations
 - Chemistry Acids And Bases D Answers User Reviews and Ratings
 - Chemistry Acids And Bases D Answers and Bestseller Lists
5. Accessing Chemistry Acids And Bases D Answers Free and Paid eBooks
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 - Chemistry Acids And Bases D Answers eBook Subscription Services
 - Chemistry Acids And Bases D Answers Budget-Friendly Options
 6. Navigating Chemistry Acids And Bases D Answers eBook Formats
 - ePub, PDF, MOBI, and More
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 - Chemistry Acids And Bases D Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Acids And Bases D Answers
 - Highlighting and Note-Taking Chemistry Acids And Bases D Answers
 - Interactive Elements Chemistry Acids And Bases D Answers
 8. Staying Engaged with Chemistry Acids And Bases D Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Acids And Bases D Answers
 9. Balancing eBooks and Physical Books Chemistry Acids And Bases D Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Acids And Bases D Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry Acids And Bases D Answers
 - Setting Reading Goals Chemistry Acids And Bases D Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry Acids And Bases D Answers

- Fact-Checking eBook Content of Chemistry Acids And Bases D Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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