

11 • Chemistry of Carbon

- A molecule named propene has the formula:
 - C_3H_6
 - C_2H_2
 - C_4H_8
 - C_3H_8
 - C_3H_4

Carbon
- What is the name of a molecule with the formula, C_5H_{10} ?
 - pentene
 - pentane
 - butyne
 - butene

Carbon
- A molecule that fits the pattern, C_nH_{2n+2} , is a(n)
 - alkane
 - alkene
 - cyclic compound
 - alkyne
 - need more info
- The general formula of the alkyne series is:
 - C_nH_{2n}
 - C_nH_{2n-4}
 - C_nH_{2n-2}
 - C_nH_{2n-6}
- How many π bonds are present in propyne?
 - 2
 - 1
 - 3
 - 4

C-C≡C
- The hydrocarbon series that has a double bond between carbon atoms is the
 - alkene family
 - alkane family
 - alkyne family
 - alkadiene family
- Organic chemistry is the study of
 - compounds of carbon
 - nonmetals
 - once-living things
 - compounds produced by living things

(HISTORIC IDEA)

PRACTICE TEST

- In octene, the hybridization(s) of carbon is(are):
 - sp^3 and sp^2
 - sp^2 only
 - sp^3 only
 - sp , sp^2 and sp^3

-C-C=C-C-C-C-C-
- The properties of carboxylic acids include all of the following except:
 - they ionize completely in water solution
 - they can be neutralized by bases *TRUE*
 - the molecule has some polarity *TRUE*
 - they react with alcohols to form esters *TRUE*
- Dimethylbenzene is also known as xylene. 1,3-dimethylbenzene would be the same as...
 - meta-xylene
 - ortho-xylene
 - para-xylene
 - none of these
- The compound
 

is classified as a

 - ketone
 - alcohol
 - ester
 - aldehyde
 - acid
- What hybridization is utilized by the oxygen atom in the carbonyl group above?
 - sp^2
 - sp^3
 - sp

*-C=O: SN=3
sp²*

Chemistry Chapter 13 Test Answer Key

**Barbara M. Castleberry, Mary E.
Lunz, Patricia A. Tanabe**



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the two general kinds of nuclear reactions are nuclear decay reactions and nuclear transmutation reactions in a nuclear decay reaction also called radioactive decay an unstable nucleus emits radiation and is transformed into the nucleus of one or more other elements

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study with quizlet and memorize flashcards containing terms like attractive force that acts between all nuclear particles that are extremely close together such as protons and neutrons in a nucleus stability of a nucleus depends on the band of stability and more

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jan 10 2017 nuclear transformation nuclear transformation reactions are nuclear reactions in which the nucleus of an element called target is bombarded with an accelerated particle called projectile bomb to transform the target into a new nucleus having a new chemical physical properties the projectiles bombs can be accelerated by using

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transformations can occur as a result of nuclear decay of radioactive isotopes or neutron capture neither procedure requires a change in the atom's element an atom gets transmuted into an atom of a different element when the number of protons in it is changed

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nuclear chemistry is the subfield of chemistry dealing with radioactivity nuclear processes such as nuclear transmutation and nuclear properties it is the chemistry of radioactive elements such as the actinides radium and radon together with the chemistry associated with equipment such as nuclear reactors which are designed to perform

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unlike a normal chemical reaction that forms molecules nuclear reactions result in the transformation of one element into another this property of nuclear reactions is used in nuclear power plants to gather nuclear energy the three common types of radiations are explained below

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a nuclear reactor consists of the following a nuclear fuel a fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction the radioactive isotope is contained in tubes called fuel rods a moderator

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apr 3 2019 in order to understand the origin and character of individual radioactive emissions accompanying nuclear transformation processes we first need to answer three questions what is an unstable nucleus

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transmutation the changing of one chemical element into another transmutation involves a change in the nucleus or core of an atom and is therefore a nuclear reaction transformations may result from nuclear decay of radioactive isotopes or they can result from neutron capture

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in all accelerators the particles move in a vacuum to avoid collisions with gas molecules when neutrons are required for transmutation reactions they are usually obtained from radioactive decay reactions or from various nuclear reactions occurring in nuclear reactors

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nuclear reaction a nuclear reaction is a process in which two nuclei or a nucleus and an external subatomic particle collide to produce one or more new nuclides the nuclear reaction is a reaction that involves nucleoids the reactant nucleoid called the parent nucleoid usually transforms into a different nucleoid called the daughter

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nuclear transmutation is the conversion of one chemical element or an isotope into another chemical element 1 nuclear transmutation occurs in any process where the number of protons or neutrons in the nucleus of an atom is changed a transmutation can be achieved either by nuclear reactions in which an outside particle reacts with a nucleus

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jan 10 2021 write a balanced equation for each of the following nuclear reactions bismuth 212 decays into polonium 212 beryllium 8 and a positron are produced by the decay of an unstable nucleus neptunium 239 forms from the reaction of uranium 238 with a neutron and then spontaneously converts into plutonium 239