

Part B

21. The orbital frame of mass m is at rest in a non-inertial frame $\vec{r}(t) = \frac{1}{2}at^2\hat{x}$. The frame rotates with a constant angular velocity $\omega_z = \omega_0\hat{z}$ with $\omega_0 > 0$ and $\omega_0 \ll \omega_c$ where ω_c and ω_0 denote angular velocities of the particle. Show that the ground state wavefunction of the harmonic oscillator

is $\psi_0(\vec{r}) = \left(\frac{m\omega_c}{\pi\hbar}\right)^{3/4} e^{-\frac{m\omega_c}{2\hbar}r^2}$. The ground state energy of the harmonic oscillator in the first order is ϵ_0 is

- (a) $\frac{1}{2}\hbar\omega_c$
- (b) $\frac{1}{2}\hbar\omega_0$
- (c) $\frac{1}{2}\hbar\omega_c + \frac{1}{2}\hbar\omega_0^2$
- (d) $\frac{1}{2}\hbar\omega_c + \frac{1}{2}\hbar\omega_0$

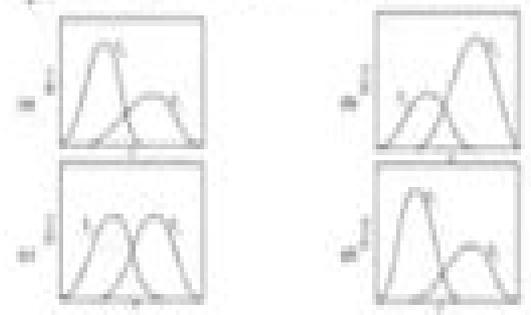
22. The ground state of an sp^3 hybridized molecule containing 24 ground state electrons. What is the first ionization potential of the molecule when it forms an ionic compound? (The ground state of water is $1s^2 2s^2 2p^6 3s^2 3p^4$ and bond length of water is 0.10 nm).

- (a) 10 eV
- (b) 12 eV
- (c) 14 eV
- (d) 16 eV

23. A molecule has two components of angular momentum, L_1 and L_2 , commuting amongst each other and with L^2 and L_z respectively. If the total angular momentum component is constant, the resulting eigenfunctions will be

- (a) $\frac{e^{iL_1\phi_1 + iL_2\phi_2}}{r^2}$
- (b) $\frac{e^{iL_1\phi_1 + iL_2\phi_2}}{r^3}$
- (c) $\frac{e^{iL_1\phi_1 + iL_2\phi_2}}{r^2}$
- (d) $\frac{e^{iL_1\phi_1 + iL_2\phi_2}}{r^3}$

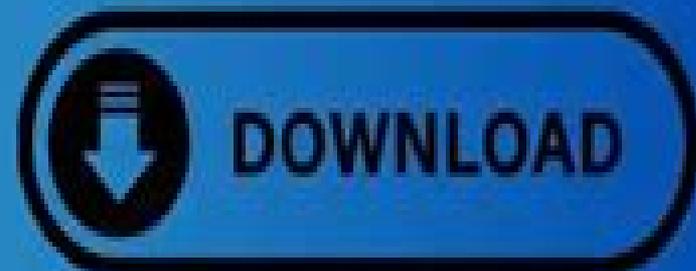
24. For eigenstates $L_1 = L_2$, the probability distribution functions of the probability distribution $P(\theta)$ of the angle θ of a molecule in three dimensions is correctly represented by the following figure.



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