

HSC Chemistry Notes - Industrial Chemistry

9.5 - 1. Industrial chemistry processes have enabled scientists to develop replacements for natural products

1. **discuss** the issues associated with shrinking world resources with regard to one identified natural product that is not a fossil fuel, **identifying** the replacement materials used and/or current research in place to find a replacement for the named material.

As the human population grows, there is increasing **demand** on natural resources; yet the supply of raw materials is **finite**. Prolonged overuse of a natural resource leads to **depletion** or even permanent loss of the resource in the future. Recycling is an important response to the problem of shrinking world resources but it cannot solve the problem. The **development of new materials is an important strategy to solve this problem**.

Synthetic Fertilisers (ammonium nitrate):

Because all living things requires nitrogen to provide nutrients to DNA and proteins, it is an essential element of air, water and soil. Nitrogen supplies are particularly important for plants and all agricultural harvests. Plants obtain their nitrogen from the soil by absorbing dissolved ammonium and nitrate ions through their roots.

Early farmers, through experimentation and observations of their crops, discovered that they could not continue to grow and re-use soils concurrently because the plants would absorb most of the ammonium ions and therefore drain the soil of nitrogen. However, the use of clovers and guano were employed by the farmers → which would replenish the nitrogen molecules in the soil. Once reserves of guano had depleted (and because clovers were an ineffective process), deposits of nitrate became available to be used in the development of nitrate rich fertilisers. Of course, however, these reserves of nitrate were finite and were used until the **Haber process** was developed for the artificial synthesis of ammonia.

The Haber process produced ammonia from nitrogen (in the air) and hydrogen (from water or natural gas). This ammonia could be oxidised to produce **nitric acid**.

- **When ammonia reacts with nitric acid, a salt called ammonium nitrate is produced**. This salt was an excellent synthetic fertiliser.



This Haber process for the synthesis of ammonia became essential for the production of synthetic fertilisers when combined with nitric acid. Not only could **NH₄NO₃** be used for the production of fertilisers so that nitrogen could be replenished in the soils - and therefore better crops - but the Haber process was vital to Germany throughout WWII in the production of ammonium nitrate to be used in explosives.

Ongoing research is conducted today by environmental and analytical chemists, to discover methods which will enhance the yield of ammonia from the Haber process (*Refer to Chemical Monitoring & Management*), however data is already determined to achieve such results from a revolutionary process.

1. **Identify data, gather and process information to identify and discuss the issues associated with the increased need for a natural resource that is not a fossil fuel and evaluate the progress currently being made to solve the problems identified**

Refer to orange dot point 1.1 - above.

2. Many industrial processes involve manipulation of equilibrium reactions

1. **explain the effect of changing the following factors on identified equilibrium reactions**

Changing equilibrium reactions will be affected according to **Le Chatelier's principle** → any change will be countered to **minimise that change**.

- **pressure**

Changing pressure will only affect an equilibrium if there is at least one gas present in a reaction chamber. When pressure is increased, the volume occupied by the gas must try to adjust to this; as such the system will shift to the side which results in the **least number of moles of each reactant/product**.

Hsc Chemistry Notes Industrial Chemistry

Padhraic Smyth



Hsc Chemistry Notes Industrial Chemistry:

This book delves into Hsc Chemistry Notes Industrial Chemistry. Hsc Chemistry Notes Industrial Chemistry is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Hsc Chemistry Notes Industrial Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Hsc Chemistry Notes Industrial Chemistry
- Chapter 2: Essential Elements of Hsc Chemistry Notes Industrial Chemistry
- Chapter 3: Hsc Chemistry Notes Industrial Chemistry in Everyday Life
- Chapter 4: Hsc Chemistry Notes Industrial Chemistry in Specific Contexts
- Chapter 5: Conclusion

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6. In chapter 5, the author will draw a conclusion about Hsc Chemistry Notes Industrial Chemistry. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Hsc Chemistry Notes Industrial Chemistry.

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1. Understanding the eBook Hsc Chemistry Notes Industrial Chemistry
 - The Rise of Digital Reading Hsc Chemistry Notes Industrial Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Chemistry Notes Industrial Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Chemistry Notes Industrial Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Chemistry Notes Industrial Chemistry
 - Personalized Recommendations
 - Hsc Chemistry Notes Industrial Chemistry User Reviews and Ratings
 - Hsc Chemistry Notes Industrial Chemistry and Bestseller Lists
5. Accessing Hsc Chemistry Notes Industrial Chemistry Free and Paid eBooks
 - Hsc Chemistry Notes Industrial Chemistry Public Domain eBooks
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6. Navigating Hsc Chemistry Notes Industrial Chemistry eBook Formats
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 - Hsc Chemistry Notes Industrial Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hsc Chemistry Notes Industrial Chemistry
 - Highlighting and Note-Taking Hsc Chemistry Notes Industrial Chemistry
 - Interactive Elements Hsc Chemistry Notes Industrial Chemistry

8. Staying Engaged with Hsc Chemistry Notes Industrial Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Chemistry Notes Industrial Chemistry
9. Balancing eBooks and Physical Books Hsc Chemistry Notes Industrial Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Chemistry Notes Industrial Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Hsc Chemistry Notes Industrial Chemistry
 - Setting Reading Goals Hsc Chemistry Notes Industrial Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hsc Chemistry Notes Industrial Chemistry
 - Fact-Checking eBook Content of Hsc Chemistry Notes Industrial Chemistry
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements
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