

Covalent Compounds & Covalent Bonding: Multiple Choice

1. What is a covalent bond?

- A) A bond formed by the transfer of electrons
- B) A bond formed by the sharing of electrons
- C) A bond formed by the sharing of protons
- D) A bond formed by the transfer of protons

ANS: B

1. What is the main characteristic of a covalent bond?

- A) Electrons are transferred from one atom to another.
- B) Electrons are shared between atoms.
- C) Protons are shared between atoms.
- D) Neutrons are transferred from one atom to another.

ANS: B

1. Which of the following is a molecular compound?

- A) calcium oxide
- B) strontium chloride
- C) dinitrogen monoxide
- D) ammonium chloride

ANS: C

1. What type of bond is likely to form between the elements molybdenum, Mo and chlorine, Cl based on electronegativity?

- A) Non polar covalent
- B) Polar covalent
- C) Ionic
- D) None of the above

ANS: B

1. What type of bond is likely to form between the elements molybdenum, Mo and nitrogen, N based on electronegativity?

- A) Non polar covalent
- B) Polar covalent
- C) Ionic
- D) None of the above

ANS: B

1. Which of the following compounds is covalent?

- A) NaCl
- B) MgO
- C) H₂O
- D) CaCl₂

ANS: C

1. Which of the following pairs of elements is most likely to form a covalent bond?

Covalent Bonding Molecular Compounds Multiple Choice

H. Stephen Stoker, Danny V. White



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 - Chapter 5: Conclusion
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