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web may 28 2020 write the balanced equation for the reaction and calculate how much  $\text{sio}_2$  is required to produce 3.00 kg of sic answer  $\text{ce sio}_2$  3c  $\rightarrow$  sic 2co

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web google classroom balance the following chemical equation  $\text{Mg(OH)}_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$  note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem  $7 \text{ x } 4 = 1 \text{ x } x \text{ y } y \theta \pi 8 \text{ 5 } 2 \text{ 0 } 9 \text{ 6}$

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web a common type of stoichiometric relationship is the mole ratio which relates the amounts in moles of any two substances in a chemical reaction we can write a mole ratio for a pair of substances by looking at the coefficients in front of each species in

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