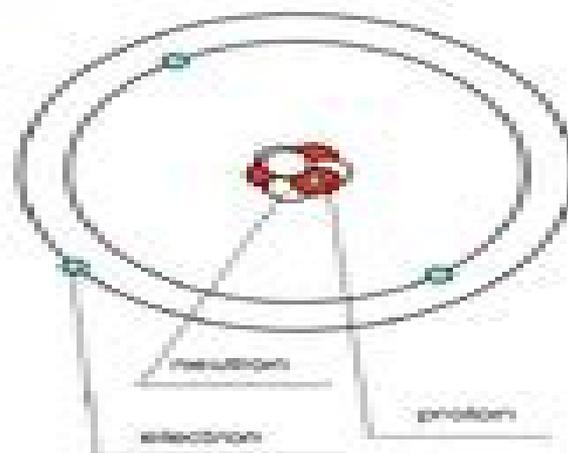


Answer Key

1. **PROTONS** are found in the center of the atom in a region called the **NUCLEUS**. (slide 7)
2. It is the number of **protons** that determine what **ELEMENT** an atom will be. (slide 8)
3. The number of **protons** is called the **ATOMIC NUMBER**. (like an ID Number) (slide 9)
4. The **atomic number** is a whole number found with the **SYMBOL** of the element. (slide 10)
5. **Protons** have a mass of **1 amu** and are one of the heavier pieces of an atom. (slide 11)
6. **Protons** have a **+1** charge. (slide 12)
7. **NEUTRONS** are the other type of particle found in the center of the atom (nucleus). (slide 13)
8. **Neutrons** have **ZERO** charge and have a mass of **1 amu**. (slide 14-15)
9. **ELECTRONS** are the smallest of the subatomic particles, found outside the nucleus. (slide 16)
10. **Electrons** have virtually no significant **MASS**. (slide 16)
11. **Electrons** have a **-1** charge. (slide 17)
12. **Electrons** can **MOVE** between atoms or be **SHARED** between atoms. (slide 17)
13. If an atom is **NEUTRAL**, it will have the same number of electrons as protons. (slide 20)
14. **Electrons** are involved in **all CHEMICAL REACTIONS**. (slide 21)

PARTICLE	MASS	CHARGE	LOCATION	ROLE
PROTON	1	+1	Nucleus	Identity of Element
NEUTRON	1	0	Nucleus	Holds Nucleus Together
ELECTRON	0	-1	Outside Nucleus	Reactivity



Atomic Structure Guided Notes Answers Boypic

Suzanne Slade



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Atomic Structure Teacher's Guide Tina Youngman,2017 The Structure of Atoms Suzanne Slade,2006-08-15
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