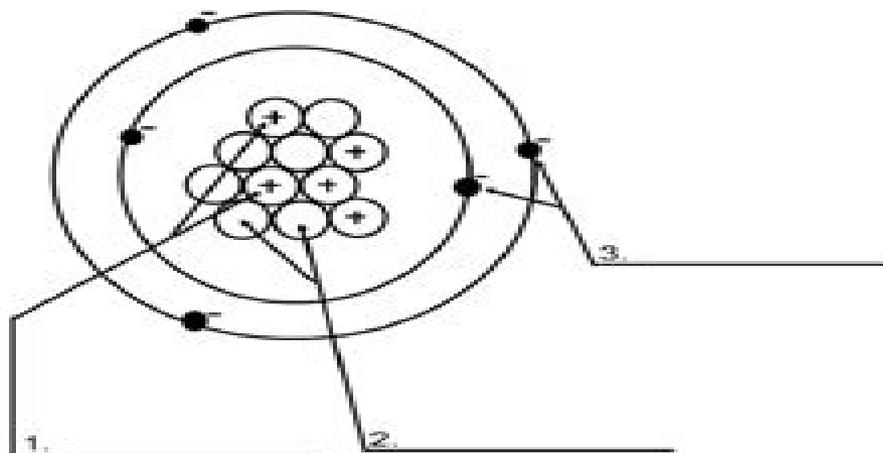


## Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?
5. What type of charge does a neutron have?
6. What type of charge does an electron have?
7. Which two subatomic particles are located in the nucleus of an atom?
8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
9. What is the atomic number of the atom in the diagram above?
10. What is the atomic mass/mass number of the atom in the diagram above?
11. How many protons are in the nucleus of an atom with an atomic number of 15?
12. How many electrons are in the nucleus of an atom with an atomic number of 20?
13. How many neutrons are in the nucleus of an atom with an atomic number of 25?  
(use Periodic Table for mass)
14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?

# Atomic Structure Guided Practice Problem Answers

**Wenbin Ji**



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*Survival Guide to General Chemistry* Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

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Student's Guide to Chemistry; a Modern Introduction David Brooks, 1974

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