

Atomic Structure Worksheet

Name _____

- ① Sodium - 23
- ② Na
- ③ P + N

Fill in the blanks for the elements in this chart. For the purposes of this chart, round all atomic masses to the nearest whole number.

$N = \text{mass\#} - P$ $= \#P$ (in a neutral atom)

Element	Atomic Symbol	Atomic Number	Number of Protons	Mass Number	Number of Neutrons	Number of Electrons
Lithium-9	Li	3	3	9	6	3
Carbon-12	C	6	6	12	6	6
Chlorine-35	Cl	17	17	35	18	17
Silver-108	Ag	47	47	108	61	47
Lead-207	Pb	82	82	207	125	82
Calcium-40	Ca	20	20	40	20	20
Potassium-39	K	19	19	39	20	19
Radium-226	Ra	88	88	226	138	88
Aluminum-27	Al	13	13	27	14	13
Krypton-84	Kr	36	36	84	48	36
Iron-56	Fe	26	26	56	30	26
Bromine-80	Br	35	35	80	45	35
Element	Atomic Symbol	Atomic Number	Number of Protons	Atomic Mass	Number of Neutrons	Number of Electrons

PT

PT

same

atom

Atomic Structure The Periodic Table Answer Key

M Walker



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Table of Contents Atomic Structure The Periodic Table Answer Key

1. Understanding the eBook Atomic Structure The Periodic Table Answer Key
 - The Rise of Digital Reading Atomic Structure The Periodic Table Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure The Periodic Table Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure The Periodic Table Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure The Periodic Table Answer Key
 - Personalized Recommendations
 - Atomic Structure The Periodic Table Answer Key User Reviews and Ratings
 - Atomic Structure The Periodic Table Answer Key and Bestseller Lists
5. Accessing Atomic Structure The Periodic Table Answer Key Free and Paid eBooks

- Atomic Structure The Periodic Table Answer Key Public Domain eBooks
 - Atomic Structure The Periodic Table Answer Key eBook Subscription Services
 - Atomic Structure The Periodic Table Answer Key Budget-Friendly Options
6. Navigating Atomic Structure The Periodic Table Answer Key eBook Formats
- ePub, PDF, MOBI, and More
 - Atomic Structure The Periodic Table Answer Key Compatibility with Devices
 - Atomic Structure The Periodic Table Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Atomic Structure The Periodic Table Answer Key
 - Highlighting and Note-Taking Atomic Structure The Periodic Table Answer Key
 - Interactive Elements Atomic Structure The Periodic Table Answer Key
8. Staying Engaged with Atomic Structure The Periodic Table Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure The Periodic Table Answer Key
9. Balancing eBooks and Physical Books Atomic Structure The Periodic Table Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure The Periodic Table Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure The Periodic Table Answer Key
- Setting Reading Goals Atomic Structure The Periodic Table Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Structure The Periodic Table Answer Key
- Fact-Checking eBook Content of Atomic Structure The Periodic Table Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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