

ACS EXAM (2024 version) Practice Questions and Answers

A 20.0 g sample of mercury(II) oxide (HgO , $M = 216.6$)

is heated strongly, causing it to decompose to metallic Hg and O_2 gas. What volume of O_2 gas is produced

(measured at STP)?

(A) 1.03 L

(B) 2.07 L

(C) 4.14 L

(D) 14.0 L - $\checkmark\checkmark$ Correct-A

When 30.0 mL of 0.10 M AgNO_3 is added to 30.0 mL of

0.10 M NaCl , aqueous NaNO_3 and solid AgCl are

formed. How much solid AgCl is produced?

(A) 0.0030 mol

(B) 0.0060 mol

(C) 0.030 mol

(D) 0.060 mol - $\checkmark\checkmark$ Correct-A

How much $\text{Sr}(\text{OH})_2 \cdot 8 \text{H}_2\text{O}$ ($M = 265.76$) is needed to

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