

ABG Background Basics

Fluid-Balance Systems: Maintains homeostasis of the body's intracellular and extracellular fluid-solvent acids and bases. Control of the fluid's maintenance of the integrity of the cellular membranes. A change in pH of ± 0.4 either direction can be fatal.

Chemical Buffer Systems
 (The most common)
 Bicarbonate Buffer Systems

Physiologic Processes
 Respiratory System
 Renal System

Bicarbonate Buffer Systems work to stabilize and the acid-base balance rapidly changed.

Respiratory System works to stabilize the oxygen and decrease carbon dioxide by decreasing ventilation. The lungs can decrease CO2 by decreasing ventilation.

Renal System works to stabilize bicarbonate levels. It can add or remove bicarbonate from the blood.

Other than stabilizing agents, the three electrolytes or minerals: Sodium, Potassium, and Calcium.

Disseminated Toxicity, Compartment, or Toxic Compartment

Normal ABG Values

pH	7.35-7.45	Alkalosis
CO2	35-45	Respiratory
HCO3	22-28	Metabolic

ABG Interpretation

There are many ways to interpret data, but one of the easiest is to work through it using a "30-30" table.

Fill up your table with the first row labeled: **Acidosis, Normal, and Alkalosis**. With the pH results, we will fill the second and third rows. Keep a column of normal values close by so a chart about would you're learned them.

Acidosis	Normal	Alkalosis

pH		
Acidosis	7.35-7.45	Alkalosis
CO2		
Acidosis	35-45	Alkalosis
HCO3		
Acidosis	22-28	Alkalosis

Normal ABG

pH Results:
 pH 7.41
 CO2 39
 HCO3 23

Acidosis	Normal	Alkalosis

A normal ABG will have all 3 values in a straight line below the Normal column.

ABG INTERPRETATION

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ABG Practice Questions

15

pH Results:
 pH 7.45
 CO2 35
 HCO3 25

16

pH Results:
 pH 7.38
 CO2 38
 HCO3 26

17

pH Results:
 pH 7.35
 CO2 40
 HCO3 27

18

pH Results:
 pH 7.32
 CO2 42
 HCO3 28

19

pH Results:
 pH 7.30
 CO2 44
 HCO3 30

Uncompensated Respiratory Acidosis

pH Results:
 pH 7.35
 CO2 48
 HCO3 24

We know it is **respiratory** in nature because the pH and the CO2 are in the same column.

We know it is **acidotic** because the majority of numbers are in the **Acidosis** column.

We know it is **uncompensated** because it has one half of an arrow above it.

Uncompensated Metabolic Acidosis

pH Results:
 pH 7.29
 CO2 37
 HCO3 18

We know it is **metabolic** in nature because the pH and the HCO3 are in the same column.

We know it is **acidotic** because the majority of numbers are in the **Acidosis** column.

We know it is **uncompensated** because it has one half of an arrow above it.

15 Pages

Abg Interpretation Practice Case Studies With Answers

BM King



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