

PHYSICAL SCIENCE – INTERMOLECULAR FORCES OF ATTRACTION

Activity 1: Identify the most probable intermolecular force of attraction in the following:

1. $\text{KCl} \cdots \text{KCl}$ -Ionic Bond
2. $\text{Na}_2\text{S} \cdots \text{Na}_2\text{S}$ -Ionic Bond
3. $\text{HF} \cdots \text{HF}$ -Hydrogen Bond
4. $\text{CH}_3\text{OH} \cdots \text{CH}_3\text{OH}$ -Hydrogen Bond
5. $\text{H}_2 \cdots \text{H}_2$ -Hydrogen Bond

Activity 2:

- What is common to H_2O , NH_3 , and HF which makes them boil at an unusually high temperature?

They all have hydrogen bonds. The boiling points of H_2O , NH_3 , and HF are much higher than those of other molecules of similar size and weight.

Activity 3:

Hydrogen Bonding

1. Why is water a liquid at room temperature?
The hydrogen bonds between water molecules are strong enough to hold the molecules together, but not strong enough to hold them in a fixed position. This allows the molecules to move around, but stay close together. This is why water is a liquid at room temperature.
2. How is hydrogen bonding in water?
The oxygen atom in a water molecule is more electronegative than the hydrogen atoms. This means that the oxygen atom has a partial negative charge, and the hydrogen atoms have a partial positive charge. This creates a dipole moment in the water molecule. The partial positive charge on one hydrogen atom is attracted to the partial negative charge on the oxygen atom of another water molecule. This attraction is called a hydrogen bond.

Physical Science Paper 2 Intermolecular Forces Investigation

CO Houle



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Proceedings of the Royal Society. Section A, Mathematical and Physical Science Royal Society (Great Britain),1908
Science Abstracts ,1921 **Research in Education** ,1972 *The Chemical News : and Journal of Physical Science*
,1894 Resources in Education ,1972-10 *Soviet Physics* ,1957 **Nature** ,1894 **Nature** Sir Norman Lockyer,1922
Colloid Chemistry Jerome Alexander,1926 *Scientific and Technical Aerospace Reports* ,1976 Lists citations with
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Journal of the Physical Society of Japan Nihon Butsuri Gakkai,2006 **Science Abstracts** ,1963 **The Electrical**
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