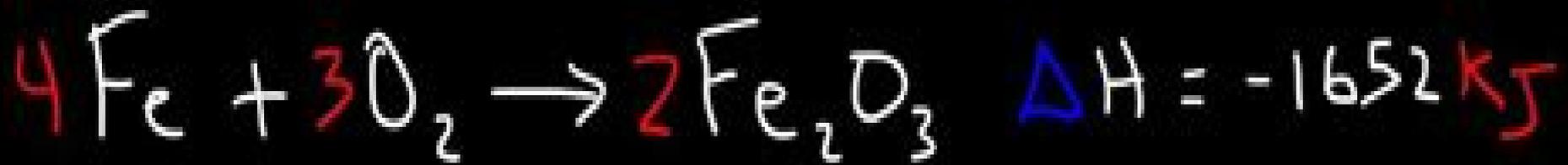


Thermochemical Equations



24 g $\text{O}_2 \rightarrow$ kJ of heat

$$\frac{24\text{g O}_2}{1} \times \frac{1\text{mol O}_2}{32\text{g O}_2} \times \frac{-1652\text{kJ}}{3\text{mol O}_2} =$$

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