

# Thermodynamics I

Final exam

Thermal physics

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## QUESTION NO. 1 (20 points)

- What is meant by an ideal gas?
  - Carbon dioxide gas is contained within a piston cylinder device initially at  $20^\circ\text{C}$  and  $0.24\text{m}^3/\text{kg}$ . It undergoes an isothermal process to state 2, where the specific volume is  $0.12\text{ m}^3/\text{kg}$ . It is then expanded at constant pressure to state 3, where the specific volume is  $0.36\text{m}^3/\text{kg}$ . Determine the initial pressure and the final pressure and temperature. Calculate the amount of work done and heat transfer through each process, in  $\text{J}/\text{kg}$ .
- 

## QUESTION NO. 2 (20points)

- A gas mixture with the partial pressure of its components given as follows:

Mixture component	Partial pressure kPa
	100
$\text{O}_2$	80
$\text{H}_2\text{O}$	20

Determine for this mixture:

- The apparent molecular weight ( $M_m$ ).
  - The gas constant ( $R_m$ ).
  - The specific heat at constant pressure ( $c_{pm}$ ).
  - The specific heat at constant volume ( $c_{vm}$ ).
  - The specific heat ratio ( $\gamma_m$ ).
- Propanol ( $\text{C}_3\text{H}_7\text{OH}$ ) is burned with 50 percent excess air. Write the balanced reaction equation for complete combustion and determine the air-to-fuel ratio, on mass basis.
- 

## QUESTION NO. 3 (15 points)

- Explain the components of the steam power plant and apply the first law of thermodynamics on the whole cycle as well as on every component.
- Steam flows steadily through an adiabatic turbine. The inlet conditions of the steam are  $10\text{ MPa}$ ,  $450^\circ\text{C}$ , and  $80\text{ m/s}$  and the exit conditions are  $10\text{ kPa}$ , 92 percent quality, and  $5\text{m/s}$ . The mass flow rate of the steam is  $12\text{ kg/s}$ . Determine, the power output and the turbine inlet and exit cross sectional area.

# Thermodynamics Final Exam

**Victor M. Corman**



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## **Table of Contents Thermodynamics Final Exam**

1. Understanding the eBook Thermodynamics Final Exam
  - The Rise of Digital Reading Thermodynamics Final Exam
  - Advantages of eBooks Over Traditional Books
2. Identifying Thermodynamics Final Exam
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Thermodynamics Final Exam
  - User-Friendly Interface
4. Exploring eBook Recommendations from Thermodynamics Final Exam
  - Personalized Recommendations
  - Thermodynamics Final Exam User Reviews and Ratings
  - Thermodynamics Final Exam and Bestseller Lists
5. Accessing Thermodynamics Final Exam Free and Paid eBooks
  - Thermodynamics Final Exam Public Domain eBooks
  - Thermodynamics Final Exam eBook Subscription Services
  - Thermodynamics Final Exam Budget-Friendly Options

6. Navigating Thermodynamics Final Exam eBook Formats
  - ePub, PDF, MOBI, and More
  - Thermodynamics Final Exam Compatibility with Devices
  - Thermodynamics Final Exam Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Thermodynamics Final Exam
  - Highlighting and Note-Taking Thermodynamics Final Exam
  - Interactive Elements Thermodynamics Final Exam
8. Staying Engaged with Thermodynamics Final Exam
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Thermodynamics Final Exam
9. Balancing eBooks and Physical Books Thermodynamics Final Exam
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Thermodynamics Final Exam
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Thermodynamics Final Exam
  - Setting Reading Goals Thermodynamics Final Exam
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Thermodynamics Final Exam
  - Fact-Checking eBook Content of Thermodynamics Final Exam
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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