

Chem 331, Exam 1

Friday, September 21, 2007 (100 points total)

$$R = 8.3145 \text{ J K}^{-1} \text{ mol}^{-1} = 0.082058 \text{ L atm K}^{-1} \text{ mol}^{-1}$$

$$1 \text{ atm} = 101.325 \text{ kPa} = 760 \text{ Torr} = 1.01325 \text{ bar} \quad (1 \text{ bar} = 10^5 \text{ Pa}); \quad 1 \text{ L} = 1 \text{ dm}^3$$

$$\mu_{J-T} = \left(\frac{\partial T}{\partial P} \right)_H, \quad \left(\frac{T_f}{T_i} \right) = \left(\frac{V_f}{V_i} \right)^{\gamma - \frac{C_v}{C_p}}, \quad w = -nRT \ln \frac{V_f}{V_i}$$

Problems (20 pts each) Show all work

- 1.) Two moles of an ideal monatomic gas initially at a pressure of 1.00 bar and 25°C is compressed adiabatically and irreversibly by application of a constant external pressure of 15.00 bar, until the final pressure of the gas reaches 10.00 bar. Assume the gas has molar heat capacity $\bar{C}_v = 3R/2$ and is independent of temperature.

- (a) Calculate the heat absorbed by the gas.
 (b) Calculate the final temperature of the gas.

a) $q = 0$ (adiabatic)

b) $\Delta U = w \Rightarrow \int_{T_i}^{T_f} n \bar{C}_v dT = \int_{V_i}^{V_f} -P_{\text{ext}} dV$ (since ideal gas)

$$n \bar{C}_v (T_f - T_i) = -P_{\text{ext}} (V_f - V_i)$$

$$= -P_{\text{ext}} \left[\frac{nRT_f}{P_f} - \frac{nRT_i}{P_i} \right]$$

rearrange:

$$T_f = T_i \frac{n \bar{C}_v + nR \frac{P_{\text{ext}}}{P_i}}{n \bar{C}_v + nR \frac{P_{\text{ext}}}{P_f}}$$

$$= 298 \frac{(2 \cdot 3R/2) + (2)(R)(15)}{(2 \cdot 3R/2) + (2)(R)(10)}$$

$$= 1639 \text{ K}$$

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James Eric Wynfield Rhodes



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Concise Physical Chemistry (for Intermediate Students) James Eric Wynfield Rhodes, 1952 *The Hidden Curriculum—Faculty-Made Tests in Science* Sheila Tobias, Jacqueline Raphael, 2013-06-29 This resource manual for college level science instructors reevaluates the role of testing in their curricula and describes innovative techniques pioneered by other teachers part I examines the effects of the following on lower division courses changes in exam content format and environment revisions in grading practices student response colleague reaction the sharing of new practices with other interested professionals and more The book includes a comprehensive introduction faculty composed narratives commentaries by well known science educators and a visual index to 100 more refined innovations

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