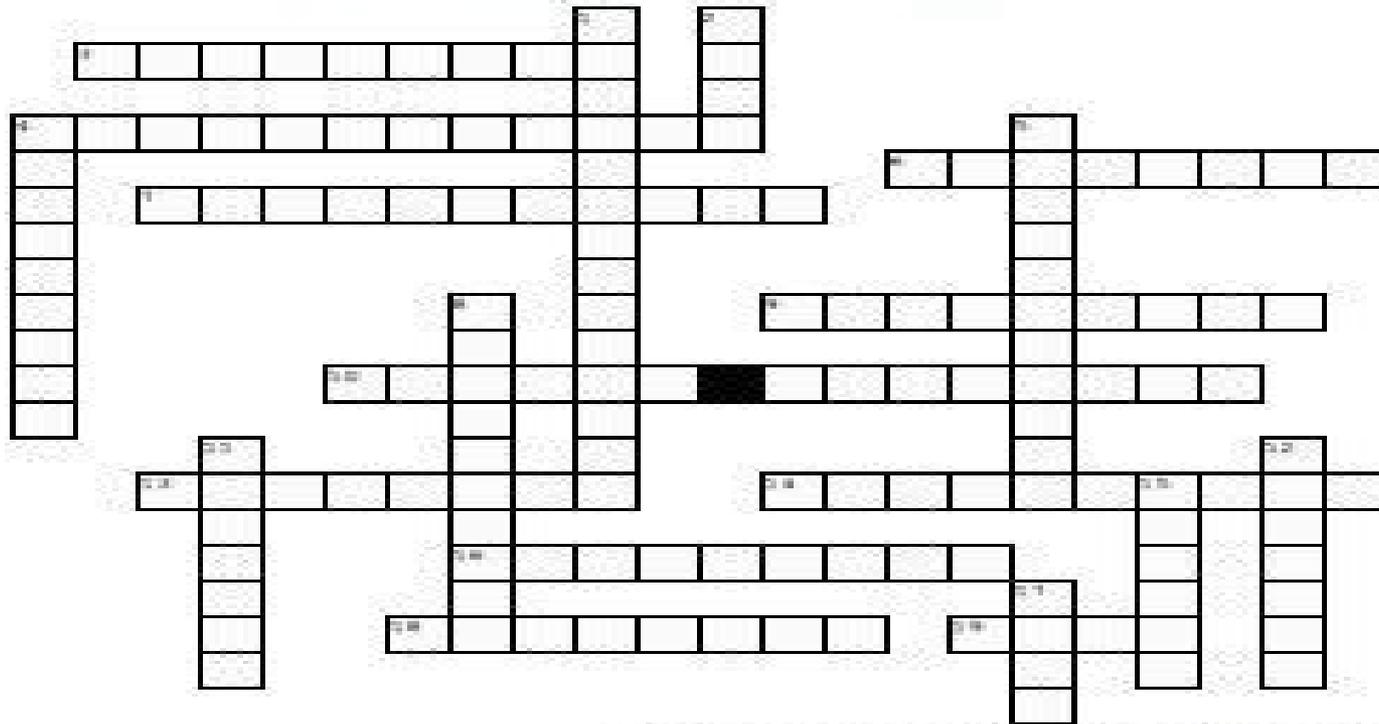


Acids, Bases, and Solutions

Complete the Crossword Puzzle Below



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Across

- The strongest bases release this ion in water.
- Describes a solution where a lot of solute is dissolved in the water.
- Solute lowers the _____ point of a solvent.
- Solution remains this way if you can continue to dissolve more solute.
- Solution becomes this when so much solute is added that no more dissolves.
- A solution where only a little solute is dissolved in the water.
- Is a well-mixed mixture.
- Is a mixture in which particles can be seen and easily separated by settling or filtration.
- A compound that changes color when in contact with an acid or a base.
- Most acids release this ion in water.
- Any ionic compound made from a neutralization of an acid with a base.

Down

- This type of acid-base reaction produces water and a salt.
- Is a substance that tastes sour, reacts with metals, and turns blue litmus paper red.
- Property of acids that describes their ability to "eat away" at other materials.
- One factor that affects solubility (hint: can be measured in degrees Celsius).
- Is a measure of how well a solute can dissolve in a solvent at a given temperature.
- Solute raises the _____ point of a solvent.
- The type of _____ also affects solubility, whether it's polar or non-polar.
- The part of a solution that is present in a solution in a smaller amount and dissolved by the solvent.
- Is a substance that tastes bitter, feels slippery, and turns red litmus paper blue.

Physical Science Acid Base And Solutions Crossword Puzzle Answers

Zhe Xu



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