

MOLE TO MOLE STOICHIOMETRY WORKSHEET

Name : _____

Date : _____

1. Using the equation



a. Find the number of moles of Al used to produce 9.9 moles of H_2 .

b. Find the number of moles of HCl used to produce 1.11 moles of AlCl_3 .

c. Find the number of moles of AlCl_3 produced along with 0.0757 moles of H_2 .

2. The reaction between lithium and fluorine produces lithium fluoride. How many moles of lithium fluoride can be produced with 2.17 mol of fluorine?



3. In the reaction of copper (II) oxide breaking down to form copper and oxygen, how much copper is produced if 0.663 moles of oxygen is given off?



4. Fermentation is a complex chemical process of making wine by converting glucose into ethanol and carbon dioxide:



If 17.5 moles of ethanol were produced, how many moles of glucose were there initially?

5. Consider the reaction:



Calculate the moles of HCl needed to react completely with 8.25 moles of zinc.

Moles And Stoichiometry Practice Problems Answers

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Stokes**



Moles And Stoichiometry Practice Problems Answers:

Unveiling the Energy of Verbal Beauty: An Emotional Sojourn through **Moles And Stoichiometry Practice Problems Answers**

In a global inundated with monitors and the cacophony of fast connection, the profound power and emotional resonance of verbal art usually diminish in to obscurity, eclipsed by the regular barrage of noise and distractions. Yet, situated within the musical pages of **Moles And Stoichiometry Practice Problems Answers**, a fascinating function of literary splendor that impulses with natural feelings, lies an wonderful trip waiting to be embarked upon. Composed by a virtuoso wordsmith, this interesting opus instructions viewers on an emotional odyssey, delicately exposing the latent potential and profound affect stuck within the elaborate internet of language. Within the heart-wrenching expanse of the evocative analysis, we can embark upon an introspective exploration of the book is main subjects, dissect its captivating publishing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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