



Communication Satisfaction In The Hospitality Industry

A

Kathleen Armour



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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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from one species to another the species that loses electrons is said to be oxidized while the species that gains electrons is said to be reduced

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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2e^{-}$ reduction $\text{Fe}^{2+} + 2e^{-} \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni} + 3 \text{ Ag}$ oxidation reduction $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb} + \text{Cu}$ oxidation reduction 26 O_2

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 $5 \text{ CO} + \text{I}_2 \rightarrow 5 \text{ CO}_2 + \text{I}_2$

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