

CONFORMATIONAL ANALYSIS – PRACTICE EXERCISES

10. Draw a Newman projection of the most stable conformation of 2-methylpropane.

11. The structures below are



- A) ethane.
- B) conformational isomers.
- C) structural isomers.
- D) both B and C.

12. Define the term *conformation*.

13. Draw a *staircase* molecule along the C_2-C_3 bond and provide a *Newman* projection of the lowest energy conformation.

14. Provide a representation of the *gauche* conformer of butane.

15. Among the butane conformers, which occur at energy minima on a graph of potential energy versus dihedral angle?

- A) gauche only.
- B) eclipsed and totally eclipsed.
- C) anti and anti.
- D) eclipsed only.
- E) anti only.

16. Which of the following substituents exhibits the greatest negative heat of combustion per CH_3 group?

- A) cyclohexane B) cyclopentane C) cyclobutane D) cyclopropane E) ethylpropane

17. Which of the following correctly ranks the substituents in order of increasing the heat per methylene (CH_2) group?

- A) cyclohexane > cyclopentane > cyclobutane > cyclopropane
- B) cyclohexane > cyclopentane > cyclobutane > ethylpropane
- C) cyclohexane > cyclopentane > cyclobutane > cyclopropane
- D) cyclohexane > cyclopentane > cyclobutane > cyclopropane
- E) cyclohexane > cyclopentane > cyclobutane > cyclopropane

18. Describe the number of angles which will be formed when propan-2-ol is cyclohexane.

Conformational Analysis Practice Exercises

A. Niederwieser, Gyorgy Pataki



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Survival Guide to Organic Chemistry Patrick E. McMahon, Bohdan B. Khomtchouk, Claes Wahlestedt, 2016-12-19 The Survival Guide to Organic Chemistry Bridging the Gap from General Chemistry enables organic chemistry students to bridge the gap between general chemistry and organic chemistry It makes sense of the myriad of in depth concepts of organic chemistry without overwhelming them in the necessary detail often given in a complete organic chemistry text Here the topics covered span the entire standard organic chemistry curriculum The authors describe subjects which require further explanation offer alternate viewpoints for understanding and provide hands on practical problems and solutions to help master the material This text ultimately allows students to apply key ideas from their general chemistry curriculum to key concepts in organic chemistry Key Features Reviews key general chemistry concepts and techniques adapted for application to important organic principles Provides practical guidance to help students make the notoriously well known and arduous transition from general chemistry to organic chemistry Explains organic concepts and reaction mechanisms generally expanding the focus on how to understand each step from a more intuitive viewpoint Covers concepts that need further explanation as well as those that summarize and emphasize key ideas or skills necessary in this field An added bonus is help with organizing principles to make sense of a wide range of similar reactions and mechanisms Implements a user friendly process to achieve the end result of problem solving Covers organic chemistry I and II concepts at the level and depth of a standard ACS organic chemistry curriculum features practice problems and solutions to help master the material including an extensive and comprehensive bank of practice exams with solutions

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