

TYPES OF CHEMICAL REACTIONS LAB PART#1

I. Purpose: To view the actual chemical reactions, write the correct balanced chemical equation, and type of chemical reaction.

Connection to Business: Businesses utilize many types of chemical reactions to produce, maintain, and sterilize their product or to prevent pollution.

II. Materials: burner, test tube, metal tongs (only...no rubber ends), microspatula, evaporating dish, wood splints

Chemicals: copper (wire), magnesium (ribbon), copper (II) carbonate

SAFETY GOGGLES!

III. Procedure: Part #1

1. Obtain materials.
2. Use the fine sandpaper to clean a piece of copper wire until shiny.
3. Set up a Bunsen burner.
4. Using tongs, hold the wire in the hottest part of the burner flame for 1-2 minutes. Record your observation of the wire's appearance afterwards.
5. Place an evaporating dish near the Bunsen burner and repeat step #4 except place the remaining magnesium into the evaporating dish.
(WARNING: AVOID LOOKING DIRECTLY AT THE FLAME)
6. Place 2 heaping microspatulas of copper (II) carbonate into a clean, dry test tube.
7. Using a test tube holder, hold the test tube (at a 45° angle pointed away from people) over the Bunsen burner flame for 3 minutes or until completely black.
(Allow this to cool slightly before placing into the wood test tube holder.)
8. Light a wood splint and place inside the test tube. Record your observations.
9. Dispose of all the metals and burnt powders in a zip-lock bag and place in the trash.
10. Clean and return all lab equipment.

IV. Data/Questions: Part #1 Data Table

Sample	Appearance Before Reaction	Appearance After Reaction
1. Copper	<i>Answer: shiny, reddish brown</i>	<i>Answer: black, powder coating</i>
2. Magnesium	<i>Answer: silver, tinsel</i>	<i>Answer: black ashes, bright light</i>
3. Mercury (II) Oxide (will not perform experiment)	<i>Reddish powder</i>	<i>Liquid mercury formed; flame burned brighter when placed in the test tube</i>
4. Copper (II) Carbonate	<i>Answer: green powder</i>	<i>Answer: decomposes into gas CO₂ and black residue</i>

Lab Report Types Of Chemical Reactions Answers

Daniel R. Palleros



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The Student's Lab Companion John W. Lehman, 2004 For undergraduate or

graduate students taking organic chemistry lab Ideal for professors who write their own lab experiments or would like custom labs but need a source for lab operations and safety information Using a practical how to approach The Student s Companion describes all of the laboratory operations that are most often used in a typical organic chemistry course It provides enough practical information to help students learn the necessary lab techniques and know how to handle problems as they arise plus just enough theory to help students understand how and why the techniques work as they do [Abstracts of Studies in Education at the Pennsylvania State College](#) Pennsylvania State College. School of Education,1936

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