

**Chapter 1**  
**Properties of Matter**

**Properties of Matter**

**QUESTIONS**

Are you able distinguish, how these questions are asked?

- Which one(s) among the following is/are physical property?
- Which one(s) among the following is/are chemical property?

**What are Physical Properties?**

Such as the colour, the odour, the melting point, the boiling point, the density, the mass, the volume, the shape, etc.



A physical property is a characteristic that can be observed without changing the identity of the substance. Most of the properties of an object can be observed. For example, you can observe an object's color, shape, size, or weight. In general, the properties of an object are determined by the particles of the substance. The shape depends on **it**.

Many of the physical properties of materials that make up objects can be measured. These properties include density, hardness, magnetism, and the ability to conduct heat and electricity.

Some physical properties depend on how much of a material you have. In other words, a large amount of a material may have different properties than a smaller amount of the material. Other physical properties do not depend on how much of the material you have. They show that no matter how large or small a sample of material is, the particular property is always the same.

**LEARNING OBJECTIVE**

Identify the physical and chemical properties of substances and distinguish between physical and chemical changes.

**LEARNING OBJECTIVE**

Identify the physical and chemical properties of substances.

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**LEARNING OBJECTIVE**

Identify the physical and chemical properties of substances.

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# Chapter 2 Properties Matter Wordwise Answers

## Mtpkitore

**Lingsheng Yao**



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