

## Chapter Test B

### Chapter: Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers in full sentences with units.

- The molar volume of an ideal gas at STP is  $22.7 \text{ L mol}^{-1}$ .  
 a. gas  
 b. pressure  
 c. molar volume  
 d. volume
- The volume of a gas is  $1.0 \text{ L}$  at  $1.0 \text{ atm}$  and  $273 \text{ K}$ .  
 a. volume  
 b. pressure  
 c. standard molar volume  
 d. number of particles
- An unknown gas has a density of  $1.96 \text{ g L}^{-1}$  at STP.  
 a. gas  
 b. density in grams  
 c. molar  
 d. volume
- The molar mass of a gas is  $44.0 \text{ g mol}^{-1}$ .  
 a. molar mass  
 b. molar volume  
 c. molar mass  
 d. molar volume
- The molar mass of a gas is  $44.0 \text{ g mol}^{-1}$ .  
 a. molar mass  
 b. molar volume  
 c. molar mass  
 d. molar volume
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 d. molar volume
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 a. molar mass  
 b. molar volume  
 c. molar mass  
 d. molar volume

# Chapter Test B Chemical Equations And Reactions Answers

**Julia Schneider**



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