

Name :

Date :

Chemistry Atomic Structure Worksheet

① For the atomic structure shown below right, answer the following questions.

a) Which element is the atom shown above? _____

b) What is the chemical symbol of the element? _____

c) What is the mass of the atom shown above? _____

d) How many valence electrons are shown in the atom above? _____



② Which subatomic particle did J.J. Thomson discover in order to create his plum pudding model of the atom? _____

③ The nucleus was discovered by which scientist:

a) Thomson

b) Dalton

c) Rutherford

d) Democritus

④ What is the mass of Carbon-14: _____

⑤ What is the number of protons in Carbon-14: _____

⑥ What is the number of neutrons in Carbon-14: _____

⑦ What is the number of protons in the radioactive Uranium-238: _____

⑧ What is the number of neutrons in Sulfur-33: _____

⑨ Neutral atoms of the same element can differ in

a) Mass number

b) Atomic number

c) Number of protons and electrons

⑩ The basic principles of atomic theory were first conceived by

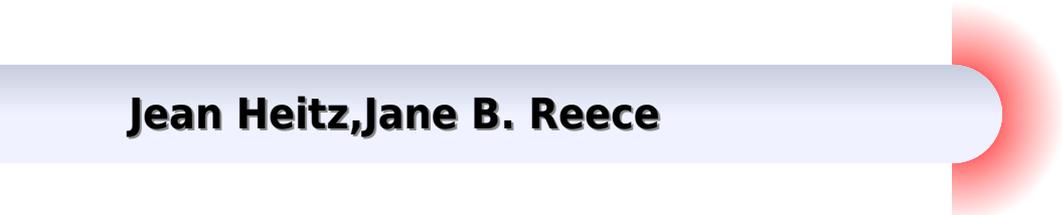
a) Avogadro

b) Bohr

c) Dalton

Pearson Chemistry Atomic Structure Test Answers

Jean Heitz, Jane B. Reece



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