

9/2/16

S. Sai Krishna

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ClassmateThermochemistry

$$Q = \Delta E - W$$

$$Q = \Delta E - (-P\Delta V)$$

$$\boxed{Q = \Delta E + P\Delta V}$$

- At constant Volume.

$$Q = \Delta E$$

- At const. Pressure.

$$Q_p = \Delta E + P\Delta V$$

$$\boxed{H = \Delta E + P\Delta V}$$

$$\boxed{H = \Delta E + \Delta nRT}$$

$\Delta n =$ no. of moles of Prod
- no. of moles of react.

→ For ΔH .

- If ΔH is -ve. reaction is exothermic.
- If ΔH is +ve. reaction is endothermic.

Q. The heat of combustion of ethane at 17°C and const volume is -322.1 kJ . Calculate the heat of combustion at constant pressure.



Engineering Chemistry Full Notes

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The Journal of Industrial and Engineering Chemistry ,1913 Industrial & Engineering Chemistry ,1925
Journal of Industrial and Engineering Chemistry ,1922 **Mining and Chemical Engineering Review** ,1908
Chemical Engineering and Mining Review ,1909 Chemical & Metallurgical Engineering ,1918 **Engineering News** ,1914 *An Index of U.S. Voluntary Engineering Standards. Supplement* William J. Slattery,1972 **An Index of U.S. Voluntary Engineering Standards, Supplement 2** William J. Slattery,1975 Technics ,1904 **The United States Catalog** George Flavel Danforth, Marion Effie Potter,1900 The Elements of Electrical Engineering Tyson Sewell,1903 **Federal Register** ,1973-12 *Marine Engineering Log* ,1912 *The Workman's Manual of Engineering Drawing* John Maxton,1891 **A rudimentary treatise on land and engineering surveying** Thomas Baker (Civil engineer),1900 **Lockwood's dictionary of terms used in ... mechanical engineering, ed. by a foreman pattern-maker [J.G. Horner].** Joseph Gregory Horner,1902 **"Verbal" Notes and Sketches for Marine Engineer Officers** John William Major Sothern,1947 **The Journal of Education** ,1892 **International Marine Engineering** ,1912

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