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# Incoder Inductive Angle Encoder Inductive Encoder

**Y Pai**

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structure hybridization geometry hinh trigonal 8 tetrahedral  $sp^3$  pyramidal  $nh_3$  h fb f

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web the pcl<sub>3</sub> lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh<sub>3</sub> chlorine cl

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periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and  
lone pairs surrounding the central atom and lastly identify the structure geometry