

	A	B	C	D	E	F	G
Molecule or Polyatomic Ion	No. of Valence Electrons	Lewis Structure		Central Atom Geometry	Bond Angles	Bond Polarity	Molecular Dipole or Nonpolar
$C_2H_3Cl_2$ * 2 central atoms	24			trigonal pyramidal & tetrahedral			
HCN	10	$H-C \equiv N:$		tetrahedral			
$CH_2O$	12			tetrahedral			
$OF_2$	20			Bent			
HBr	8	$H-Br:$					
$PCl_3$	26			trigonal pyramidal			
$NO_3^-$	24			tetrahedral			

\* More than one possible Lewis structure can be drawn. See questions 1, 2.

\*\* More than one possible Lewis structure can be drawn. See question 2.

# Lewis Structures Molecular Model Lab Answers

**R Sandford**



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