

# POGIL Activity – Gas Variables

## (Teacher Notes)

- Assessment Questions and Responses:

### Assessment Questions

- Select the equation below that gives the correct relationship between the internal pressure of a gas and the volume of a flexible container.
  - $P = kV$
  - $P = kV$
  - $V = kP$
  - $k = p/V$
- Explain, on the molecular level, the change in internal pressure of a gas sample in a nonflexible container when the temperature is raised.
- Explain, on the molecular level, the change in volume in a flexible container when more gas is added.

### Assessment Target Responses

- a.*
- When the temperature of a gas is increased, the average kinetic energy of the molecules increases. This makes the gas molecules hit harder and more often, which increases the pressure.*

# Pogil Gas Variables Answers

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