

Assessment

Chapter Test A**Chapter: States of Matter**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas particles are considered to have no definite
- mass.
 - volume.
 - density.
 - energy.
- _____ 2. Under which conditions does a real gas behave very much like an ideal gas?
- high temperature and low pressure
 - high temperature and high pressure
 - low temperature and high pressure
 - low temperature and low pressure
- _____ 3. If a gas and a liquid are at the same temperature and pressure, diffusion occurs much faster in the gas because
- there are more elastic collisions between the particles in a gas.
 - gases are more compressible.
 - the particles move faster in a gas and there is a greater distance between them.
 - gas molecules are in continuous motion.
- _____ 4. If the particles in a sample of matter are attracted to each other but can move past each other easily, the matter is a
- solid.
 - liquid.
 - gas.
 - plasma.
- _____ 5. According to the equation for the kinetic energy of a moving object, which gas would have the lowest average velocity, assuming all gases are at the same temperature?
- O_2
 - H_2
 - H_2O
 - O_3

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traditional high stakes testing towards more meaningful ways of evaluating student learning New chapter that highlights and explains key elements of the Common Core State Standards for English Language Arts and shows how the incorporation of critical ELA instruction into the social studies curriculum can foster more ambitious teaching and learning Real classroom narratives that introduce each chapter and provide in depth access to teaching and learning contexts Practical curriculum and resource suggestions for the social studies classroom End of chapter summaries and annotated teaching resources

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In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Assessment Chapter Test B States Of Matter**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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