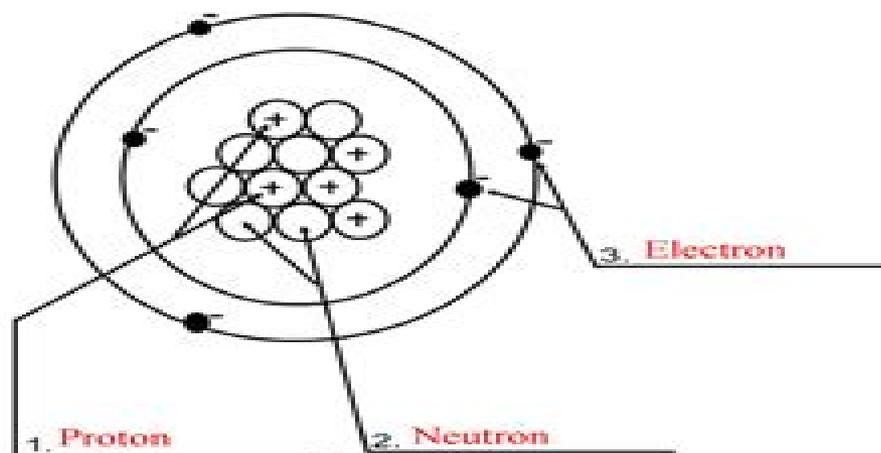


## Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?  
**Positive (+1)**

5. What type of charge does a neutron have?  
**Neutral (0)**

6. What type of charge does an electron have?  
**Negative (-1)**

7. Which two subatomic particles are located in the nucleus of an atom?  
**Protons and neutrons**

8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?  
**35**

9. What is the atomic number of the atom in the diagram above?  
**5**

10. What is the atomic mass/mass number of the atom in the diagram above?  
**(11+11=22) \*protons and neutrons added together) 22 is the atomic mass/mass number**

11. How many protons are in the nucleus of an atom with an atomic number of 15?  
**15 \*number of protons is the same as the atomic number**

12. How many electrons are in the nucleus of an atom with an atomic number of 20?  
**20**

13. How many neutrons are in the nucleus of an atom with an atomic number of 25?  
(use Periodic Table for mass)  
**30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30**

14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?  
**Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.**

15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?  
**Neutrons= 11 (atomic mass: 36 - 25 :atomic number)**

# Atomic Structure Chapter 4 Worksheet Answers

**Sebastian Brünink**



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### **Table of Contents Atomic Structure Chapter 4 Worksheet Answers**

1. Understanding the eBook Atomic Structure Chapter 4 Worksheet Answers
  - The Rise of Digital Reading Atomic Structure Chapter 4 Worksheet Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Chapter 4 Worksheet Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atomic Structure Chapter 4 Worksheet Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Chapter 4 Worksheet Answers
  - Personalized Recommendations
  - Atomic Structure Chapter 4 Worksheet Answers User Reviews and Ratings
  - Atomic Structure Chapter 4 Worksheet Answers and Bestseller Lists
5. Accessing Atomic Structure Chapter 4 Worksheet Answers Free and Paid eBooks
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  - Atomic Structure Chapter 4 Worksheet Answers eBook Subscription Services

- Atomic Structure Chapter 4 Worksheet Answers Budget-Friendly Options
- 6. Navigating Atomic Structure Chapter 4 Worksheet Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Atomic Structure Chapter 4 Worksheet Answers Compatibility with Devices
  - Atomic Structure Chapter 4 Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Atomic Structure Chapter 4 Worksheet Answers
  - Highlighting and Note-Taking Atomic Structure Chapter 4 Worksheet Answers
  - Interactive Elements Atomic Structure Chapter 4 Worksheet Answers
- 8. Staying Engaged with Atomic Structure Chapter 4 Worksheet Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Atomic Structure Chapter 4 Worksheet Answers
- 9. Balancing eBooks and Physical Books Atomic Structure Chapter 4 Worksheet Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Atomic Structure Chapter 4 Worksheet Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Chapter 4 Worksheet Answers
  - Setting Reading Goals Atomic Structure Chapter 4 Worksheet Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Chapter 4 Worksheet Answers
  - Fact-Checking eBook Content of Atomic Structure Chapter 4 Worksheet Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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