



Designation: D 4373 – 96

Standard Test Method for Calcium Carbonate Content of Soils¹

This standard is issued under the fixed designation D 4373; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last approval. A superscript symbol (n) indicates an editorial change since the last revision or approval.

1. Scope

1.1 This test method covers the quantitative determination of the calcium carbonate (CaCO_3) content of soils. It is a gasometric method that utilizes a simple portable apparatus. The test method is quickly performed for soils containing calcium carbonate.

Note 1—The presence of dolomite (CaMgCO_3) and reducing minerals such as sulfide and sulfite in the soil will interfere with the determination of the amount of CaCO_3 present. Therefore, this test method is an approximate method that determines the calcium carbonate equivalent.

1.2 The values stated in SI units are to be regarded as the standard.

1.3 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use. For specific precaution statements, see Section 6.

2. Summary of Test Method

2.1 The calcium carbonate content of soil is determined by treating a 1-g dried soil specimen with hydrochloric acid (HCl) in an enclosed reactor vessel. Carbon dioxide gas is evolved during the reaction between the acid and carbonate fraction of the specimen. The resulting pressure generated in the closed reactor is proportional (see Fig. 1) to the carbonate content of the specimen. This pressure is measured with a Bourdon tube pressure gage, or equivalent pressure-measuring device, that is precalibrated with reagent grade calcium carbonate.

3. Significance and Use

3.1 This test method is used to determine the presence and quantity of calcium carbonate in a soil specimen. Calcium carbonates are known cementing agents, are water soluble at $\text{pH} < 7$, and are soft on the Mohs' scale compared to other soil minerals.

¹This test method is under the jurisdiction of ASTM Committee D 18 on Soil and Rock, and is the direct responsibility of Subcommittee D18.11 on Marine Geotechnics.

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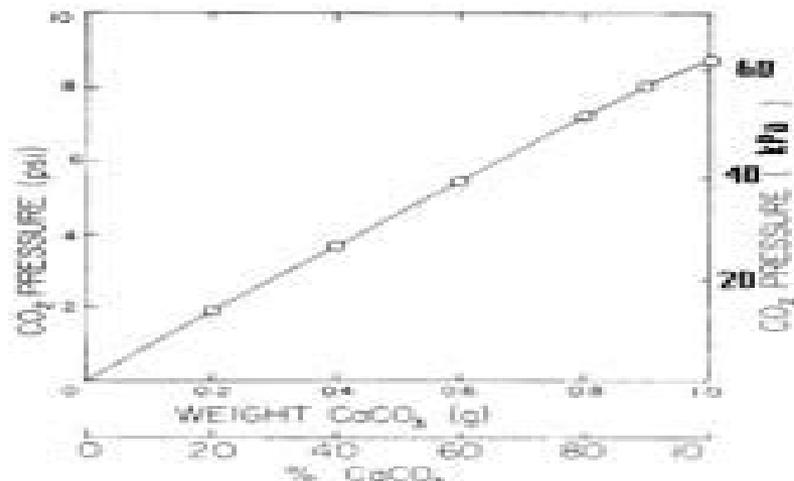


FIG. 1 Typical Calibration Curve for 0.374L Test Cell and 10 psi (69 kPa) Pressure Gage

4. Apparatus

4.1 **Rapid Carbonate Analyzer**—A schematic drawing of the rapid carbonate analyzer is shown in Fig. 2. The basic components of this apparatus include:

4.1.1 **Reaction Cylinder**, with threaded cap and O-ring seal to enclose the cylinder. A clear plastic cylinder allows viewing of effervescent reaction.

4.1.2 **Pressure Gage**, 10 ± 0.1 psi (70 ± 0.7 kPa).

4.1.3 **Container**, of clear plastic with a bail handle to hold 20 mL of acid.

4.1.4 **Pressure Relief Valve**, for safe release of CO_2 gas pressure.

4.2 **Balance**, with a sensitivity of 0.01 g.

5. Reagents

5.1 **Purity of Reagents**—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that

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