

7.3

BONDING IN METALS

Section Review

Objectives

- 1. Identify metallic bonding characteristics
- 2. Describe the delocalized electron model
- 3. Explain the properties of metals

Vocabulary

- metallic bond
- metal

Part A: Completion

Use the appropriate word or phrase to complete each statement. Write your answer in the space provided. Use a dictionary if you need to check the meaning of a word.

1. The delocalized electron model is a model of metallic bonding.
2. Electrons are the particles that are delocalized in a metal.
3. Electrons are the particles that are delocalized in a metal.
4. Electrons are the particles that are delocalized in a metal.
5. Electrons are the particles that are delocalized in a metal.
6. Electrons are the particles that are delocalized in a metal.
7. Electrons are the particles that are delocalized in a metal.
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10. Electrons are the particles that are delocalized in a metal.

11. Electrons are the particles that are delocalized in a metal.
12. Electrons are the particles that are delocalized in a metal.
13. Electrons are the particles that are delocalized in a metal.
14. Electrons are the particles that are delocalized in a metal.
15. Electrons are the particles that are delocalized in a metal.
16. Electrons are the particles that are delocalized in a metal.
17. Electrons are the particles that are delocalized in a metal.
18. Electrons are the particles that are delocalized in a metal.
19. Electrons are the particles that are delocalized in a metal.
20. Electrons are the particles that are delocalized in a metal.

Part B: True/False

Write **True** or **False** in the space provided for each statement.

1. True Metals are good conductors of electricity.
2. True Metals are good conductors of heat.
3. True Metals are good conductors of electricity.
4. True Metals are good conductors of heat.
5. True Metals are good conductors of electricity.
6. True Metals are good conductors of heat.
7. True Metals are good conductors of electricity.
8. True Metals are good conductors of heat.
9. True Metals are good conductors of electricity.
10. True Metals are good conductors of heat.

Bonding In Metals Section Review Answers Key

VM Jensen



Bonding In Metals Section Review Answers Key:

Decoding **Bonding In Metals Section Review Answers Key**: Revealing the Captivating Potential of Verbal Expression

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