

CHAPTER 7 TEST FOR ORGANIC CHEMISTRY QUESTIONS WITH COMPLETE SOLUTIONS

an example of chemical changes is _____

- A. water changing to ice
- B. ice melting to water
- C. the oxidation of haircolor products
- D. temporary haircolor - Answer-C. the oxidation of haircolor products

_____ refers to either the loss of oxygen or the addition of hydrogen.

- A. application
- B. reduction
- C. reaction
- D. oxidation - Answer-B. reduction

when using a permanent wave neutralizer _____

- A. H₂O₂ loses oxygen
- B. the waving solution is oxidized
- C. haircolor gains oxygen from H₂O₂
- D. hair is oxidized by removing hydrogen - Answer-D. hair is oxidized by removing hydrogen

when oxygen is combined with a substances, the substances is _____

- A. oxidized
- B. reduced
- C. United
- D. blended - Answer-A. oxidized

oxidation and reduction always occur simultaneously and are referred to as a _____ reaction

- A. rapid oxidation
- B. endothermic
- C. redox
- D. endothermic - Answer-C. redox

what type of chemical reaction requires the absorption of energy or heat from an external source for the reaction to actually occur?

- A. endothermic
- B. acid-alkali neutralization
- C. exothermic
- D. redox - Answer-A. endothermic

which of the following is an example of a pure substances?

- A. concrete
- B. Aluminum foil
- C. powder

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