

Name _____

Multiple Choice (10 points each)

Identify the letter of the choice that best completes the statement or answers the question.

- E 1. For a certain process at 27°C, $\Delta G = +210.6$ kJ and $\Delta H = -168.2$ kJ. What is the entropy change for this process at this temperature? Express your answer in the form, $\Delta S =$ _____ J/K.
- a. +628.3 J/K
 b. -141.3 J/K
 c. +141.3 J/K
 d. 1.26×10^3 J/K
 e. -1.26×10^3 J/K

Answer: $\Delta G = \Delta H - T\Delta S$ $T = 27^\circ\text{C} + 273 = 300\text{K}$

$$\Delta S = \frac{\Delta H - \Delta G}{T} = \frac{(-168.2\text{kJ}) - (+210.6\text{kJ})}{300\text{K}} = -\frac{1.26\text{kJ}}{\text{K}}$$

$$\left(\frac{-1.26\text{kJ}}{\text{K}}\right)\left(\frac{10^3\text{J}}{\text{kJ}}\right) = \boxed{\frac{-1.26 \times 10^3\text{J}}{\text{K}}}$$

- A 2. Which response includes all of the following substances that have $\Delta H_f^\circ = 0$, and no other substances?

- I. HCl(g)
 II. Na(s)
 III. HCl(aq)
 IV. F₂(g)

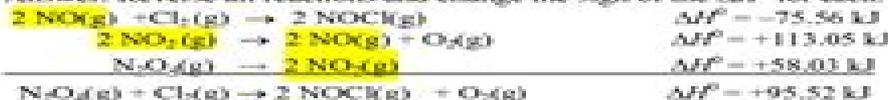
- a. II and IV
 b. II
 c. I and II
 d. I, II, III, and IV
 e. I, II, and IV

Answer: All elements in their natural standard state have a $\Delta H_f^\circ = 0$.

- A 3. From the following data at 25°C,

Calculate ΔH° at 25°C for the reaction below.

- a. +95.52 kJ
 b. +299 kJ
 c. -186.8 kJ
 d. -85.52 kJ
 e. -246.65 kJ

Answer: Reverse all reactions and change the sign of the ΔH° for each.

Thermodynamics Problems And Answers

Max Fogiel



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Thermodynamics Problem Solver The Editors of REA, Ralph Pike, 2013-01-01 REA's Thermodynamics Problem Solver Each Problem Solver is an insightful and essential study and solution guide chock full of clear concise problem solving gems Answers to all of your questions can be found in one convenient source from one of the most trusted names in reference solution guides More useful more practical and more informative these study aids are the best review books and textbook companions available They're perfect for undergraduate and graduate studies This highly useful reference provides thorough coverage of pressure work and heat energy entropy first and second laws ideal gas processes vapor refrigeration cycles mixtures and solutions For students in engineering physics and chemistry

Thermodynamics Problem Solving in Physical Chemistry Kathleen E. Murphy, 2020-03-23 Thermodynamics Problem Solving in Physical Chemistry Study Guide and Map is an innovative and unique workbook that guides physical chemistry students through the decision making process to assess a problem situation create appropriate solutions and gain confidence through practice solving physical chemistry problems The workbook includes six major sections with 20-30 solved problems in each section that span from easy single objective questions to difficult multistep analysis problems Each section of the workbook contains key points that highlight major features of the topic to remind students of what they need to apply to solve problems in the topic area Key Features Provides instructor access to a visual map depicting how all equations used in thermodynamics are connected and how they are derived from the three major energy laws Acts as a guide in deriving the correct solution to a problem Illustrates the questions students should ask themselves about the critical features of the concepts to solve problems in physical chemistry Can be used as a stand alone product for review of Thermodynamics questions for major tests

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Solutions Manual for Quanta, Matter and Change Peter Atkins, Julio dePaula, Ron Friedman, 2008-12-15

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carrying systems leading to an understanding of Maxwell's equations. Students can experiment with both advanced graphing and numerical techniques. Systems requirements are 80386/80486 PC or compatibles, Windows 3.1 or higher, 3.5 disk drive, 4 MB of RAM and 4 MB of disk space.

US Solutions Manual to Accompany Elements of Physical Chemistry 7e David Smith, 2017-09-28. The Solutions Manual to Accompany Elements of Physical Chemistry 7th edition contains full worked solutions to all end of chapter discussion questions and exercises featured in the book. The manual provides helpful comments and friendly advice to aid understanding. It is also a valuable resource for any lecturer who wishes to use the extensive selection of exercises featured in the text to support either formative or summative assessment and wants labour saving ready access to the full solutions to these questions.

The Thermodynamics of Electrical Phenomena in Metals, and A Condensed Collection of Thermodynamic Formulas Percy Williams Bridgman, 1961.

Commonly Asked Questions in Thermodynamics Marc J. Assael, William A. Wakeham, Anthony R. H. Goodwin, Stefan Will, Michael Stamatoudis, 2011-03-10. Have you ever had a question that keeps persisting and for which you cannot find a clear answer? Is the question seemingly so simple that the problem is glossed over in most resources or skipped entirely? CRC Press Taylor and Francis is pleased to introduce *Commonly Asked Questions in Thermodynamics*, the first in a new series of books that address the questions that frequently arise in today's major scientific and technical disciplines. Designed for a wide audience from students and researchers to practicing professionals in related areas, the books are organized in a user-friendly Question Answer format. Presented questions become increasingly specific throughout the book with clear and concise answers as well as illustrations, diagrams and tables where helpful. Thermodynamics is a core discipline associated with the theoretical principles and practical applications underlying almost every area of science from nanoscale biochemical engineering to astrophysics. Highlighting chemical thermodynamics in particular, this book is written in an easy-to-understand style and provides a wealth of fundamental information, simple illustrations and extensive references for further research and collection of specific data. Designed for an audience that ranges from undergraduate students to scientists and engineers at the forefront of research, this indispensable guide presents clear explanations for topics with wide applicability. It reflects the fact that very often the most common questions are also the most profound.

Physical Chemistry Student Solutions Manual Charles Trapp, Peter Atkins, Marshall Cady, 2006-08-11. Change 21.

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Classical and Quantum Thermal Physics R. Prasad, 2016-11-02. Discusses the interactions of heat energy and matter.

Thermal Physics and Statistical Mechanics S. K. Roy, 2001. This book emphasises the development of problem solving skills in undergraduate science and engineering students. The book provides more than 350 solved examples with complete step-by-step solutions as well as around 100 practice problems with answers. Also explains the basic theory, principles, equations and formulae for a quick understanding and review. Can serve both as a useful text and companion.

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Problems in Chemical Thermodynamics with Solutions Maka Aleksishvili,Shota Sidamonidze,2002 The methods of chemical thermodynamics are effectively used in many fields of science and technology Mastering these methods and their use in practice requires profound comprehension of the theoretical questions and acquisition of certain calculating skills This book is useful to undergraduate and graduate students in chemistry as well as chemical thermal and refrigerating technology it will also benefit specialists in all other fields who are interested in using these powerful methods in their practical activities

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