

# Lab 5: Molecular Models



Table 1: Molecular Models

| Example Atom<br>(central atom is<br>underlined>) | Number of<br>Bonds on Cen-<br>tral Atom<br>(multiple bonds<br>count as 1) | Unshared elec-<br>tron pairs on<br>central atom | Regions of<br>Electron Den-<br>sity | Molecular Geom-<br>etry   | Structure                    |
|--|---|---|-------------------------------------|---------------------------|------------------------------|
| <u>C</u> O <sub>2</sub>                          | 2   | 0   | 2                                   | Linear                    | $O=C=O$                      |
| <u>Al</u> Cl <sub>3</sub>                        | 3   | 0   | 3                                   | Trigonal Planar           | $Cl-Al-Cl$<br>$Cl$           |
| <u>O</u> H <sub>2</sub>                          | 2   | 2   | 4                                   | Bent                      | $H-O-H$<br>:<br>:            |
| <u>N</u> H <sub>3</sub>                          | 3   | 1   | 4                                   | Trigonal pyrami-<br>dal   | $H-N-H$<br>:<br>H            |
| <u>C</u> Cl <sub>4</sub>                         | 4   | 0   | 4                                   | Tetrahedral               | $Cl-C-Cl$<br>$Cl-Cl$         |
| <u>P</u> Cl <sub>5</sub>                         | 5   | 0   | 5                                   | Trigonal bi-<br>pyramidal | $Cl-Cl$<br>$Cl-P-Cl$<br>$Cl$ |
| <u>S</u> F <sub>6</sub>                          | 6   | 0   | 6                                   | Octahedral                | $F-F$<br>$F-S-F$<br>$F-F$    |

# Molecular Models Shapes Lab Answers

**Terry C. Jones**



**Molecular Models Shapes Lab Answers:**

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