

Formula	Lewis Structure	Shape	Bond Angle	Dipole (yes/no)
7 CH <sub>3</sub> OH	<p>O: 24 V: 14</p>	tetrahedron	109.5°	Yes
8 SO <sub>2</sub> Cl <sub>2</sub>	<p>O: 40 V: 32</p>	Tetrahedron	109.5°	Yes
9 OCl <sub>2</sub>	<p>O: 24 V: 20</p>	Bent	109.5°	Yes
10 SO <sub>2</sub>	<p>O: 24 V: 18</p>	Bent	120°	Yes
11 SO <sub>3</sub>	<p>O: 32 V: 24</p>	Trigonal Planar	120°	No
12 ClO <sub>3</sub> <sup>-</sup>	<p>O: 32 V: 26</p>	Trigonal Pyramid	109.5°	Yes
13 N <sub>2</sub> O	<p>O: 24 V: 16</p>	Linear	180°	Yes
O <sub>3</sub>		Bent	120°	Yes

# Molecular Geometry Lab With Answer Key

**Jiyuan Zhang**



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