

AQA A-level Biology

Condensed notes

MONOMERS & POLYMERS

- Monomers - the smaller units from which large molecules are made
- Polymers - molecules made from many monomers of identical or similar structures
- Condensation reaction - a reaction where bonds are formed and water is released
- Hydrolysis reaction - a reaction where bonds are broken and water is released

Polypeptide	Monomer	Bond
polysaccharide	monosaccharide	glycosidic bond
polyamide	amino acid	peptide bond
Protein and e.g. DNA + RNA	nucleotide	phosphodiester bond

*Small and very similar monomers that can be made of repeating units

INORGANIC IONS

- Simplest ion - an ion that doesn't contain other ions
- Can be an anion (negative) or cation (positive) charge
- Often not with negative charge
- Often not with positive charge

Ion	Role
Hydrogen ions (H⁺)	<ul style="list-style-type: none"> involved in many biological processes control the pH of a solution act as a catalyst in many biological reactions act as a cofactor in many enzymes act as a signal in many cells act as a cofactor in many enzymes act as a cofactor in many enzymes
Hydroxide ions (OH⁻)	<ul style="list-style-type: none"> involved in many biological processes control the pH of a solution act as a catalyst in many biological reactions act as a cofactor in many enzymes act as a signal in many cells act as a cofactor in many enzymes act as a cofactor in many enzymes
Calcium ions (Ca²⁺)	<ul style="list-style-type: none"> involved in many biological processes control the pH of a solution act as a catalyst in many biological reactions act as a cofactor in many enzymes act as a signal in many cells act as a cofactor in many enzymes act as a cofactor in many enzymes
Magnesium ions (Mg²⁺)	<ul style="list-style-type: none"> involved in many biological processes control the pH of a solution act as a catalyst in many biological reactions act as a cofactor in many enzymes act as a signal in many cells act as a cofactor in many enzymes act as a cofactor in many enzymes

NUCLEIC ACIDS

NUCLEOTIDE - the basic unit from which nucleic acids are made
 Polymers called **NUCLEIC ACIDS** (DNA and RNA) are made of nucleotides
 DNA and RNA are both **polynucleotides** - chains of nucleotides joined together by phosphodiester bonds



- Monomers are joined together by phosphodiester bonds - forming a continuous chain of nucleotides
- The end of sugar nucleotides is the sugar phosphate backbone
- Phosphodiester bond is formed between the phosphate group of one nucleotide and the 3' carbon of the sugar of the next nucleotide

NUCLEIC ACID - a long chain of nucleotides joined together by phosphodiester bonds
 DNA is a double helix structure
 RNA is a single helix structure
 The sugar phosphate backbone is the same in both DNA and RNA
 The nitrogenous bases are different in DNA and RNA



CARBOHYDRATES

MONOSACCHARIDE - the simplest form of carbohydrate
 Polymers called **POLYSACCHARIDES** are made of monosaccharides
 Polysaccharides are made of monosaccharides joined together by glycosidic bonds
 Polysaccharides are made of monosaccharides joined together by glycosidic bonds



- Monomers are joined together by glycosidic bonds - forming a continuous chain of monosaccharides
- The end of sugar nucleotides is the sugar phosphate backbone
- Phosphodiester bond is formed between the phosphate group of one nucleotide and the 3' carbon of the sugar of the next nucleotide

NUCLEIC ACID - a long chain of nucleotides joined together by phosphodiester bonds
 DNA is a double helix structure
 RNA is a single helix structure
 The sugar phosphate backbone is the same in both DNA and RNA
 The nitrogenous bases are different in DNA and RNA

Biological Molecules

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