

CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron III sulfate, the roman numeral tells us **C**
 - How many atoms of Fe are in one formula unit.
 - How many sulfate ions can be attached to the iron atom
 - The charge on each Fe ion
 - The total positive charge of the formula unit
- The result of changing a subscript in a correctly written chemical formula is to **c**
 - Change the number of moles represented in the formula
 - Change the charges on the other ions in the compound
 - Change the formula so that it no longer represents the compound
 - Have no effect on the formula
- Derive a generalization for when an acid name will end in the suffix **-ic** or **-ous**
If ending is -ic, the anion ended in -ate; if ending is -ous, the anion ended in -ite
- Derive a generalization for when an acid name will begin with the prefix **hydro-**.
If prefix is hydro-, the anion ended in -ide. This is most likely a monatomic ion.
- Fill in the blanks in the table below

Compound Name	Formula
Aluminum sulfide	Al_2S_3
Aluminum sulfate	$Al_2(SO_4)_3$
Lead II chloride	$PbCl_2$
Ammonium phosphite	$(NH_4)_2PO_3$
Hydroiodic acid	HCl
Dinitrogen pentoxide	N_2O_5
Copper II carbonate	$CuCO_3$
Sodium sulfide	Na_2S
Ammonium phosphate	$(NH_4)_3PO_4$
Tin IV sulfite	$Sn(SO_3)_2$
Nitrous acid	HNO_2
Magnesium perchlorate	$Mg(ClO_4)_2$
Iron II nitrate	$Fe(NO_3)_2$
Iron III nitrate	$Fe(NO_3)_3$
Cobalt II oxide	CoO
Sulfur trioxide	SO_3
Boron pentafluoride	BF_5
Tetraphosphorus octasulfide	P_4S_8
Tetraiodide nonoxide	I_4O_9

- Name the following compound and identify as molecular, ionic or acid
 - $Ca(HCO_3)_2$ **calcium bicarbonate - I**
 - $Fe(ClO_2)_3$ **Iron III chlorite - I**
 - $Fe(ClO_4)_2$ **Iron II chlorate - I**
 - N_2O_5 **dinitrogen pentoxide - M**

Chapter 9 Chemical Names Formulas Quiz Answer Key

Christian G. Meyer



Chapter 9 Chemical Names Formulas Quiz Answer Key:

Chapter 9 Chemical Names Formulas Quiz Answer Key Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter 9 Chemical Names Formulas Quiz Answer Key**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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