

## Section 10.1

### Section 10.1

Answer complete answer multiple-choice questions that are provided.

1. \_\_\_\_\_
  - a. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
  - b. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
2. \_\_\_\_\_
  - a. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
  - b. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
3. \_\_\_\_\_
  - a. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
  - b. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_

4. Given the two reactions given by the following balanced equations:



a. Balance the equation:



\_\_\_\_\_ b. What is the coefficient of  $\text{H}_2\text{O}$  in  $\text{H}_2\text{O}$ ?

\_\_\_\_\_ c. How many molecules of  $\text{H}_2\text{O}$  are produced from 100 molecules of  $\text{H}_2$  and 100 molecules of  $\text{O}_2$ ?

\_\_\_\_\_ d. What is the mass of  $\text{H}_2\text{O}$  formed from 100 g of  $\text{H}_2$  and 100 g of  $\text{O}_2$ ?

\_\_\_\_\_ e. What is the mass of  $\text{H}_2\text{O}$  formed from 100 g of  $\text{H}_2$  and 100 g of  $\text{O}_2$ ?

# Chapter 9 Section 1 Stoichiometry Answers

**RC Schank**



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