

## Chapter 6 Thermochemistry

### Homework Problems

- The enthalpy change for the fermentation (without oxygen,  $O_2$  gas) reaction of glucose to alcohol and carbon dioxide is  $-67$  kJ.
  - Write a chemical equation for the fermentation reaction and show how energy is involved in the reaction.
  - Is the reaction exothermic or endothermic?
  - Is energy, in the form of heat, absorbed or evolved as the reaction occurs?
- The specific heat capacity of graphite is  $0.71$  J/°C·g. Calculate the energy required to:
  - Raise the temperature of  $1.0$  mol graphite by  $1.0$ °C.
  - Raise the temperature of  $850$  g graphite by  $150$ °C.
  - Raise the temperature of  $75$  kg graphite from  $294$  K to  $348$  K.
- In a coffee-cup calorimeter,  $100.0$  mL of  $1.0$  M NaOH and  $100.0$  mL of  $1.0$  M HCl are mixed. Both solutions were originally at  $24.6$ °C. After the reaction, the final temperature is  $31.3$ °C. Assume that all the solutions have a density of  $1.0$  g/mL and a specific heat capacity of  $4.18$  J/g°C. Calculate the enthalpy change for the neutralization of HCl by NaOH. Assume that no heat is lost to the surroundings or to the calorimeter.
- Camphor ( $C_{15}H_{26}O$ ) has an energy of combustion of  $-5903.6$  kJ/mol. A sample of camphor with a mass of  $0.1204$  g is burned in a bomb calorimeter, the temperature increases by  $2.28$ °C. Calculate the heat capacity of the calorimeter.  
A  $0.1964$  g sample of quinone ( $C_6H_4O_2$ ) is burned in a bomb calorimeter that has a heat capacity of  $1.56$  kJ/°C. The temperature of the calorimeter increases by  $3.2$ °C. Calculate the energy of combustion of quinone per gram and per mole.

# Chapter 6 Thermochemistry Homework Problems

**G Orfield**



## **Chapter 6 Thermochemistry Homework Problems:**

## Adopting the Track of Appearance: An Emotional Symphony within **Chapter 6 Thermochemistry Homework Problems**

In some sort of used by screens and the ceaseless chatter of quick interaction, the melodic splendor and mental symphony produced by the published word usually diminish into the backdrop, eclipsed by the relentless sound and disruptions that permeate our lives. Nevertheless, set within the pages of **Chapter 6 Thermochemistry Homework Problems** a marvelous fictional value filled with natural emotions, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, that captivating masterpiece conducts readers on a mental journey, skillfully unraveling the concealed melodies and profound affect resonating within each cautiously constructed phrase. Within the depths of this moving examination, we shall investigate the book is central harmonies, analyze its enthralling writing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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