

ch 7 p 2 ch 8  
electron configuration:

ch 5/28



s, p main group elements  
d transition metal,  
(max 28)

valence electrons: outer-most electrons, electrons at highest energy level  
→ further from nucleus

electron configuration of ions

- main group → closest noble gas in brackets
- ions w/ same electron configuration → isoelectronic

effective nuclear charge

$Z_{eff}$  = nuclear charge or net attraction

$$Z_{eff} = Z - \sigma$$

attraction from nuclear charge

shielding (screening) constant

repulsion force



# Chapter 8 Electron Configuration And General Chemistry

**Guillaume Favre**



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