

1. For 2-pentyne, which of the following describes the orbital overlap of the C2—C3 sigma bond?



- A. $sp-sp$
- B. sp^2-sp^2
- C. sp^2-sp^3
- D. $p-p$
- E. sp^2-sp

Answer: A

Learning Objective: 9.1 Describe the structure of alkynes and the nature of their chemical function, explaining how alkynes are named

Difficulty: Easy

2. For 2,4-hexadiyne, which of the following describes the orbital overlap of the C3—C4 sigma bond?



- A. $sp-sp$
- B. sp^2-sp^2
- C. sp^2-sp^3
- D. $p-p$
- E. sp^2-sp

Answer: A

Learning Objective: 9.1 Describe the structure of alkynes and the nature of their chemical function, explaining how alkynes are named

Difficulty: Easy

3. The sigma bond of an alkyne is formed from the overlap of which two orbitals?

- A. sp^2-sp^3
- B. $p-p$
- C. sp^2-sp^2
- D. $s-s$
- E. $sp-sp$

Answer: E

Chapter 9 Test Chemistry

Thomas Evangelist



Chapter 9 Test Chemistry:

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The Chemical Engineer, 1910

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- Chapter 2: Essential Elements of Chapter 9 Test Chemistry
- Chapter 3: Chapter 9 Test Chemistry in Everyday Life
- Chapter 4: Chapter 9 Test Chemistry in Specific Contexts
- Chapter 5: Conclusion

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3. In chapter 2, this book will delve into the foundational concepts of Chapter 9 Test Chemistry. The second chapter will elucidate the essential principles that need to be understood to grasp Chapter 9 Test Chemistry in its entirety.

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