



## PRE-LAB QUESTIONS

- Describe the process you would go through to find the number of valence electrons for any of the main group (columns 1–8) elements in the Periodic Table.
  - I would read the group # from the periodic table. The group number is the same as the number of valence electrons.
- Although both sugar and salt are soluble in water, the way in which they dissolve is different. Explain how their dissolving process is different, and elaborate why these differences exist.
  - Salt starts off as a compound, but dissolves in ions, which always happens. Sugar, which is covalent, is bonded because they share electrons and stay together in compounds.
- Do you think 1 gram of sucrose or 1 gram of sodium chloride will melt at a higher temperature? Explain your answer.
  - 1 gram of sodium chloride will melt at a higher temperature because it is an ionic compound, while sucrose is a covalent compound. Ionic compounds generally have much higher melting points than covalent compounds.
- Using chemical nomenclature, name the following ionic compounds:
  - NaCl: Sodium Chloride
  - KBr: Potassium Bromide
  - MgO: Magnesium Oxide

# Chemical Bonds Analysis Lab Answers

**S Baum**



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