

**Balancing and Reactions Test Unit 2 SCH3U**

Name: \_\_\_\_\_

    K        T        C        A    **Assessment of Learning (Product)**  
**15        20        5        4**

Multiple Choice: Circle the most correct answer. (10K)

1. Propane gas,  $C_3H_8$ , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for  $H_2O$  is...

- A) 1    B) 2    C) 3    D) 4    E) 5    f) 6

2. State the type of reaction:  $2 Na + 2 H_2O \rightarrow 2 NaOH + H_2$

- A) synthesis                      B) decomposition              C) single displacement
- 
- D) double displacement        E) combustion                  F) no reaction

3. Which of the following types of chemical reactions produces water and an ionic compound?

- A) single displacement              B) neutralization              C) decomposition              D) synthesis

4. State the type of reaction:  $Ca(NO_3)_2 + 2 HCl \rightarrow CaCl_2 + 2 HNO_3$

- A) synthesis                      B) decomposition              C) single displacement
- 
- D) double displacement        E) combustion                  F) no reaction

5. If chlorine gas is produced by single displacement, the other halogen in the reaction must be

- A) bromine    B) iodine    C) astatine    D) fluorine

6. In what kind of reaction do two or more substances combine to form a new compound?

- A) decomposition reaction              C) double-displacement reaction
- 
- B) single displacement                  D) synthesis reaction

7. A group whose atoms (M) undergo the reaction:



- A) group 1              B) group 2              C) group 13              D) group 14              E) group 15

8. When heated, a metal carbonate decomposes into a metal oxide and

- A) carbon.                      B) carbon dioxide              C) oxygen.                      D) hydrogen.

9. Oxides of active metals, such as  $CaO$ , react with water to produce

- A) metal carbonates.              B) metal hydrides.              C) acids.                      D) metal hydroxides.

10. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals.              C) Two reactants produce two products.
- 
- B) They involve a single product.              D) Any metal replaces any other metal.

# Chemical Reactions Test Answers

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