

Chapter 9 – Stoichiometry – Chapter Review

Last page worked out for you.

Match each description on the right to the correct term on the left.

- | | | |
|----------|--|----------------------|
| <u>B</u> | 1. The substance that determines the amount of product that can be formed in a reaction. | A. Percent yield |
| <u>E</u> | 2. The amount of product that forms when a reaction is carried out in the laboratory. | B. Limiting reagent |
| <u>D</u> | 3. The calculation of quantities in chemical equations. | C. Theoretical yield |
| <u>A</u> | 4. The ratio of the actual yield to the theoretical yield expressed as a percent. | D. Stoichiometry |
| <u>F</u> | 5. The substance that is present in enough quantity to react with a limiting reagent. | E. Actual yield |
| <u>C</u> | 6. The maximum amount of products that could be formed from given amounts of reactants. | F. Excess reagent |

Multiple Choice - Choose the best answer and write its letter in the blank.

- D 7. How many moles of H_2O would be required to produce 2.5 moles of O_2 ?
- $$2 \text{H}_2\text{O} \rightarrow 2 \text{H}_2 + \text{O}_2$$
- A. 2.0 mol C. 4.0 mol
B. 2.5 mol D. 5.0 mol
- D 8. If 3.00 mol of CaCO_3 undergo decomposition to form CaO and CO_2 , how many grams of CO_2 are produced?
- $$\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$$
- A. 3.00 g C. 88.0 g
B. 44.0 g D. 132 g
- B 9. How many moles of H_2O are produced when 240 grams of CuO react?
- $$\text{CuO} + \text{H}_2 \rightarrow \text{Cu} + \text{H}_2\text{O}$$
- A. 1.0 mol C. 18 mol
B. 3.0 mol D. 54 mol

Chemistry Chapter 9 Stoichiometry Answers

Ensheng Dong



Chemistry Chapter 9 Stoichiometry Answers:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Chemistry Chapter 9 Stoichiometry Answers**

In some sort of inundated with screens and the cacophony of instant communication, the profound energy and mental resonance of verbal artistry frequently disappear into obscurity, eclipsed by the continuous barrage of sound and distractions. Yet, nestled within the lyrical pages of **Chemistry Chapter 9 Stoichiometry Answers**, a interesting function of literary brilliance that impulses with natural emotions, lies an wonderful journey waiting to be embarked upon. Penned with a virtuoso wordsmith, that mesmerizing opus books viewers on a mental odyssey, delicately revealing the latent possible and profound affect embedded within the elaborate web of language. Within the heart-wrenching expanse of this evocative evaluation, we shall embark upon an introspective exploration of the book is main themes, dissect its fascinating publishing fashion, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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