

Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
---------------------	--------------------	-----------------------

Matter is anything with (1) mass and volume. A(2) substance is a form of matter with a uniform and unchanging composition.Substances have specific, unchanging (3) properties that can be observed.Substances have both physical and chemical properties. (4) Physical

properties can be observed without changing a substance's chemical composition. Color,

hardness, and (5) density are examples. Other properties cannot be

observed without changing the composition of a substance. These are called

(6) chemical properties. An example is the tendency of iron to form

rust when exposed to air.

Label each property as either physical or chemical.

- physical 7. Chemical formula H_2O
- chemical 8. Irons gain cohesion when exposed to heat or
- chemical 9. Remains unchanged when in the presence of nitrogen
- physical 10. Crystalline
- physical 11. Solid at normal temperatures and pressures
- chemical 12. Ability to combine with another substance
- physical 13. Melting point
- physical 14. Liquid at normal temperatures and pressures
- physical 15. Boiling point is $100^\circ C$
- physical 16. Conducts electricity
- physical 17. Density is $\frac{1}{2} \frac{g}{cm^3}$

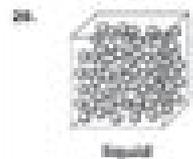
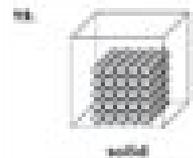
Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Section 3.1 Matter

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- true 21. All matter that we encounter in everyday life exists in one of three physical states.
- true 22. A solid has definite shape and volume.
- false 23. A liquid has a definite shape and takes on the volume of its container.
- true 24. A gas has both the shape and the volume of its container.
- false 25. The particles in a gas cannot be compressed into a smaller volume.
- false 26. Liquids tend to contract when heated.
- false 27. The particles in a solid are spaced far apart.
- false 28. The words gas and vapor can be used interchangeably.

Chemistry Matter Change Chapter Assessment Answers

Seager/Slabaugh



Chemistry Matter Change Chapter Assessment Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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[Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change Chemistry](#) John S. Phillips, Cheryl Wistrom, 2000
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[Essentials of Chemistry](#) Dennis D. Staley, Antony C. Wilbraham, 1984
Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004
The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even-numbered end-of-chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application.
[Instructor's Manual and](#)

Test Bank to Accompany Basic Concepts of Chemistry Malone,2000-04-18 *Oswaal CDS (Combined Defence Services) Chapter-wise & Topic-wise 11 Years' Solved Papers (2014-2024) General Knowledge | For 2024-25 Exam* Oswaal Editorial Board,2024-05-23 Benefits of the product 1 100% Updated with Fully Solved CDS I April 2024 Paper 2 Extensive Practice No of Questions Gen Knowledge 1200 English 1200 Mathematics 1200 3 Crisp Revision with Smart Mind Maps 4 Valuable Exam Insights with Expert Tips to crack CDS in first attempt 5 Concept Clarity with Concept based Revision Notes Detailed Explanations 6 100% Exam Readiness with 5 Years Chapter wise Trend Analysis 2019 2024 7 Exclusive Advantage of Oswaal360 Courses and Mock Papers to enrich your learning journey further **GED Test** Stuart Donnelly,2017-08-07 1 001 practice opportunities for passing the GED test Ready to take the GED test Get a head start on a high score with 1 001 GED Test Practice Questions For Dummies Inside you ll find 1 001 practice questions on all four sections of the GED test Mathematical Reasoning Science Social Studies and Reading Language Arts All of the question types and formats you ll encounter on the exam are here so you can study practice and increase your chances of scoring higher on the big day Earning a passing score on the GED test will boost your self esteem enable you to continue your education and qualify you for better paying jobs it s a win win If you re preparing for this important exam there are 1 001 opportunities in this guide to roll up your sleeves put your nose to the grindstone and get the confidence to perform your very best Includes free one year access to practice questions online Offers 1 001 GED test practice questions from easy to hard Lets you track your progress see where you need more help and create customized question sets Provides detailed step by step answers and explanations for every question Study with the book or study online or do a little of both and get ready to pass the GED test with flying colors *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Chemistry** McGraw-Hill Staff,2001-07 **Study Guide with Answers to Selected Problems** H. Stephen Stoker,Danny V. White,2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises Study Guide to accompany Basic Concepts of Chemistry, 7th Edition Leo J. Malone,2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone s Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the

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