

Multiple Choice: *Identify the letter of the choice that best completes the statement or answers the question.*

- _____ 1. What scientific (natural) law serves as the basis for balancing chemical equations by requiring that there be no observable change in the quantity of matter during a chemical reaction?
- Law of Conservation of Energy
 - Law of Multiple Proportions
 - Law of Conservation of Matter
 - Law of Definite Proportions
 - Law of Constant Composition

- _____ 2. Balance the following equation with the smallest whole number coefficients. What is the coefficient for O₂ in the balanced equation?



- 9
 - 5
 - 15
 - 6
 - 13
- _____ 3. What is the coefficient for HBr when the following equation is balanced with the smallest whole number coefficients?



- 5
 - 7
 - 8
 - 3
 - 6
- _____ 4. Balance the following equation with the smallest whole number coefficients. What is the coefficient for NH₃ in the balanced equation?



- 1
 - 3
 - 2
 - 6
 - 4
- _____ 5. Elemental phosphorus is produced from calcium phosphate in the following reaction. What is the coefficient for C when this equation is balanced with the smallest whole number coefficients?



- 10
- 3
- 1
- 6
- 4

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