

**Objectives:**

- Memorize key ions and atoms symbols and their charges
- Be able to name ionic compounds
-  Be able to name hydrates
- Be able to name acids
- Be able to name molecular compounds
-  Be able to name and write formulas for coordination compounds
-  Be able to name organic compounds
- Balance chemical equations
- Be able to count and use significant figures in calculations
- Be able to convert measurements using dimensional analysis
- Know accuracy and precision
-  Be able to calculate standard deviation and percent error

**Memorize Key ions and atoms:**

You must score a 93% to receive credit. See memory sheet for these items. Retakes will be given immediately during your study hall on Thursday.

**Naming Ionic Compounds (metal and nonmetal compounds)***Rules*

1. Name the cation
2. Use Roman Numerals if the metal is a transition metal (Exceptions: Zn, Ag) or Pb and Sn
3. Name the Anion ending in *ide* unless it is a polyatomic ion

**Examples: From Formula to Name**

- |                                   |                                      |
|-----------------------------------|--------------------------------------|
| 1. CsF                            | 6. CuBr                              |
| 2. AlCl <sub>3</sub>              | 7. CoI <sub>2</sub>                  |
| 3. LiH                            | 8. Na <sub>2</sub> SO <sub>4</sub>   |
| 4. Fe <sub>2</sub> O <sub>3</sub> | 9. Fe(NO <sub>3</sub> ) <sub>3</sub> |
| 5. PbCl <sub>2</sub>              | 10. KMnO <sub>4</sub>                |

**Examples: From Name to Formula (you must know the charges on the ions!)**

- |                          |                               |
|--------------------------|-------------------------------|
| 1. Vanadium (V) fluoride | 6. Thallium (III) perchlorate |
| 2. Gallium oxide         | 7. Barium hydrogen carbonate  |
| 3. Titanium (IV) nitrate | 8. Rubidium sulfide           |
| 4. Magnesium nitride     | 9. Sodium thiosulfate         |
| 5. Iron (II) phosphate   | 10. Strontium chloride        |

# Ap Chemistry Chapter 1 Test

**Neil D. Jespersen, Neil Jespersen, PH.  
D.**

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