

**Activity 1**

**Properties of Matter**

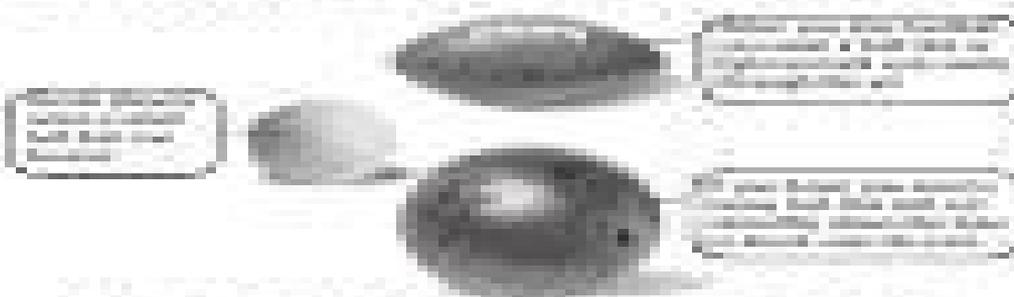
**Introduction**

Are you able to describe, how these particles are held?

- Why are solids, liquids and gaseous particles held together?
- Why are intermolecular attractive forces stronger in solids than in liquids and gases?

**What Are Physical Properties?**

Look at the table in the chapter below to find out the different physical properties of four common substances.



If physical properties of substances are observed without changing the identity of the substance, these are physical properties of an object or matter. For example, you can identify an object when its shape, colour, weight, size, etc. are changed. The properties of an object are determined by the particles of the substance. The shape depends on **it**.

Many of the physical properties of substances depend on the forces between the particles (or molecules). These properties include strength, hardness, brittleness, and the ability to conduct heat and electricity.

Some physical properties depend on how closely the particles are held. In certain cases, a large amount of a material may have different properties from smaller amounts of the material. Other physical properties do not depend on how closely the particles are held. They show that the matter has large or small amounts of particles in the particular property or always the same.

**LEARNING OBJECTIVES**

By the end of this chapter, you should be able to describe the physical properties of solids, liquids and gases. You should also be able to explain the differences between these states of matter.

**LEARNING GOALS**

By the end of this chapter, you should be able to describe the physical properties of solids, liquids and gases.

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**LEARNING OBJECTIVES**

By the end of this chapter, you should be able to describe the physical properties of solids, liquids and gases.

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# Chapter 2 Properties Matter Wordwise Answers

**Camilla Rothe**



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