

Exercises

14.1-14.3 Temperature, internal energy and states of matter

1. (a) Molecules (of the liquid) are in random / haphazard motion (AM)
Molecules (of liquid) are smaller than pollen grains.
(b) Increase the temperature (of the liquid)

2. The sum of (the random distribution of) the KE and PE of (the) molecules

3. No change in KE
Because temperature is constant (during melting)
PE of (the molecules) increases (during melting).
The internal energy increases.

14.4 Specific Heat Capacity

1. 35 000 J

2. 21 000 J/g°C

3. 0.375 kg

4. 60 s

5. a. Electrical energy supplied = Heat taken in by water
 $2.5^2 \times 20 = 25 \times 20$
 $2300 \times 2 = 2.5 \times 4200 \times 60$
 $t = 240 \text{ s [1]}$

b. The kinetic energy of the particle is increasing. [1]

6. $2 = 3 \times 60 = 180 \text{ s [1]}$

Chapter 12 Assessment Thermal Energy Answers

Lauren Gardner



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Enjoying the Tune of Phrase: An Emotional Symphony within **Chapter 12 Assessment Thermal Energy Answers**

In some sort of taken by monitors and the ceaseless chatter of fast interaction, the melodic beauty and emotional symphony developed by the published term frequently diminish in to the background, eclipsed by the persistent sound and interruptions that permeate our lives. However, situated within the pages of **Chapter 12 Assessment Thermal Energy Answers** a stunning literary prize overflowing with raw emotions, lies an immersive symphony waiting to be embraced. Crafted by a wonderful musician of language, this interesting masterpiece conducts visitors on a mental trip, well unraveling the concealed songs and profound influence resonating within each cautiously crafted phrase. Within the depths of the moving assessment, we will examine the book is key harmonies, analyze its enthralling publishing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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